

Long-Distance Proton Transfer with a Break in the Bacteriorhodopsin Active Site

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Abstract: Bacteriorhodopsin is a proton-pumping membrane protein found in the plasma membrane of the archaeon *Halobacterium salinarium*. Light-induced isomerization of the retinal chromophore from all-*trans* to 13-*cis* leads to a sequence of five conformation-coupled proton transfer steps and the net transport of one proton from the cytoplasmic to the extracellular side of the membrane. The mechanism of the long-distance proton transfer from the primary acceptor Asp85 to the extracellular proton release group during the O → bR is poorly understood. Experiments suggest that this long-distance transfer could involve a transient state [O] in which the proton resides on the intermediate carrier Asp212.¹ To assess whether the transient protonation of Asp212 participates in the deprotonation of Asp85, we performed hybrid Quantum Mechanics/Molecular Mechanics proton transfer calculations using different protein structures and with different retinal geometries and active site water molecules. The structural models were assessed by computing UV–vis excitation energies and C=O vibrational frequencies. The results indicate that a transient [O] conformer with protonated Asp212 could indeed be sampled during the long-distance proton transfer to the proton release group. Our calculations suggest that, in the starting proton transfer state O, the retinal is strongly twisted and at least three water molecules are present in the active site.

1. Introduction

The light-induced proton-pumping cycle of bacteriorhodopsin consists of five sequential proton transfer (PT) steps (Figure 1). A proton is first transferred from the chromophore's Schiff base to the nearby Asp85. The Asp85 site is electrostatically coupled to an extracellular group (denoted as the proton release group (PRG)),^{2,3} which releases a proton to the bulk once Asp85 is protonated. The Schiff base is then reprotonated from the

cytoplasmic group Asp96, which will in turn receive a proton from the cytoplasmic bulk. The last PT step (the O → bR transition in Figure 1a), and the focus of the present work, is from the primary acceptor Asp85 to the extracellular PRG over a distance of ~12 Å (Figure 1b). The mechanism of the Asp85 deprotonation and the structural details of the conformer ready for the last PT step are poorly understood. Key open questions are as follows: (i) does the PT occur in a single step, from Asp85 to the PRG, or does it involve intermediate(s), and (ii) what are the structural elements critical for PT from Asp85 to the PRG. To address these questions, we performed an extensive set of computations of the PT paths, excitation energies, and vibrational spectra.

A recent investigation in which vibrational spectra calculated with various structural models indicated that the PRG likely consists of Glu194 and Glu204 sharing a proton.⁴ The combined quantum mechanics/molecular mechanics (QM/MM) simula-

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(1) Dioumaev, A. K.; Brown, L. S.; Needleman, R.; Lanyi, J. K. *Biochemistry* **1999**, *38*, 10070–10078.

(2) Balashov, S. P.; Govindjee, R.; Imasheva, E. S.; Misra, S.; Ebrey, T. G.; Feng, Y.; Crouch, R. K.; Menick, D. R. *Biochemistry* **1995**, *34*, 8820–8834.

(3) Lanyi, J. K. *Biochim. Biophys. Acta* **1993**, *1183*, 241–261.

(4) Phatak, P.; Ghosh, N.; Yu, H.; Cui, Q.; Elstner, M. *Proc. Natl. Acad. Sci. U.S.A.* **2008**, *105*, 19672–19677.

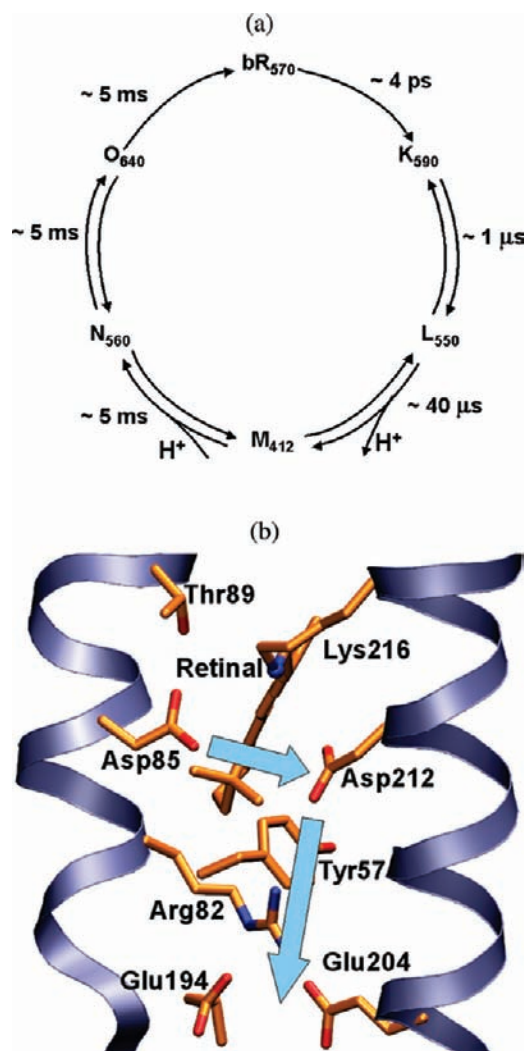


Figure 1. Bacteriorhodopsin proton-pumping cycle. (a) Schematic representation of the bacteriorhodopsin photocycle showing intermediate states characterized by their absorption maxima; (b) Location of protein groups essential for the proton transfer. The arrows indicate the proton transfer (PT) from Asp85 to the PRG during O \rightarrow bR transition. In the bacteriorhodopsin photocycle, light-activated isomerization of the retinal from all-*trans* to 13-*cis* takes place during bR \rightarrow K transition, followed by relaxation of the twisted retinal during K \rightarrow L transition. The first PT takes place from Schiff base to Asp85 along with the release of another proton to the extracellular bulk by PRG during L \rightarrow M transition. The M \rightarrow N involves reprotonation of the Schiff base from Asp96. Subsequently Asp96 is reprotonated from the cytoplasmic medium, followed by reisomerization of retinal during N \rightarrow O transition. The last PT steps takes place during O \rightarrow bR transition from Asp85 to PRG. The picture is prepared using the crystal structure of the bR resting state²⁴ and using the VMD package.¹⁰³

tions⁴ demonstrated that the excess proton binds to Glu194/Glu204 instead of the water cluster. The structure with the excess proton shared by the negatively charged Glu194 and Glu204 is stable on the nanosecond time scale of the QM/MM simulations and gives rise to a continuum band similar to that observed in previous studies.⁵ This role of Glu194/Glu204 as the PRG is consistent with previous interpretations^{6,7} but contradicts a more recent proposal that the excess proton is stored on a water cluster.^{8,9}

The path from Asp85 to Glu194/Glu204 is curved and is interrupted by the side chain of Arg82 (Figure 1b) in the light-adapted bacteriorhodopsin resting state (bR). The coupling between the deprotonation of Asp85 and protein conformational

changes is unclear. It has been suggested that the intermediate states of the second half of the photocycle (i.e., late-M, N, and O; Figure 1a) can be approximated by a single protein conformation.¹⁰ Although several crystal structures exist for M, the assignment of these structures to a specific M substate is difficult. Moreover, it has been questioned whether the crystal environment allows the full extent of protein conformational changes associated with late-M *in vivo*.^{11,12} Deprotonation of Asp85 is presumably triggered by the thermal isomerization of the retinal from 13-*cis* back to all-*trans* during the transition between the N and O intermediate states (Figure 1a). Experiments have indicated that the retinal back isomerization is coupled to protein structural rearrangements: mutations of amino acid residues close to the retinal such as Leu93 dramatically delay retinal isomerization by 250-fold.¹³

Rearrangement of water molecules is important for the coupling between Asp85 and the PRG,^{14,15} but the details of how the long-distance PT from Asp85 to the PRG is coupled to local structural rearrangements of the protein and water molecules are not clear.

Intermediate conformers distinguished by the geometry of the retinal chain and by the protonation states of key protein groups may accumulate during the O \rightarrow bR step. The rate of the deprotonation of Asp85 is influenced by the local environment around Glu204.¹⁶ The sensitivity of the O \rightarrow bR transition to temperature and pH^{17,18} indicates that this step of the photocycle is rather complex. It has been suggested¹⁶ that the thermal back isomerization of the retinal triggers the formation of a transient O intermediate in which retinal is twisted and Asp85 is deprotonated. The transient O intermediate then converts into the bR resting state after relaxation of the retinal twist. Experiments on Glu194Gln^{1,19} and Glu204Gln¹ mutant phenotypes provided further evidence for the presence of a transient O state (denoted here as [O]) between the O and the bR states. According to ref 1, the transfer of the proton from Asp85 to the PRG may involve an intermediate step [O] in which Asp85 proton had been transferred to Asp212.¹ These experiments on bacteriorhodopsin mutants^{1,19} raise the intriguing question of whether the transient [O] intermediate could also

- (5) Rammelsberg, R.; Huhn, G.; Luebben, M.; Gerwert, K. *Biochemistry* **1998**, *37*, 5001–5009.
- (6) Essen, L. O.; Siebert, R.; Lehmann, W. D.; Oesterheld, D. *Proc. Natl. Acad. Sci. U.S.A.* **1998**, *95*, 11673–11678.
- (7) Zscherp, C.; Schlesinger, R.; Tittor, J.; Oesterheld, D.; Heberle, J. *Proc. Natl. Acad. Sci. U.S.A.* **1999**, *96*, 5498–5503.
- (8) Garczarek, F.; Brown, L. S.; Lanyi, J. K.; Gerwert, K. *Proc. Natl. Acad. Sci. U.S.A.* **2005**, *102*, 3633–3638.
- (9) Garczarek, F.; Gerwert, K. *Nature (London)* **2006**, *439*, 109–112.
- (10) Subramaniam, S.; Lindahl, M.; Bullough, P.; Faruqi, A. R.; Tittor, J.; Oesterheld, D.; Brown, L.; Lanyi, J. K.; Henderson, R. *J. Mol. Biol.* **1999**, *287*, 145–161.
- (11) Sass, H. J.; Schachowa, I. W.; Rapp, G.; Koch, M. H. J.; Oesterheld, D.; Dencher, N. A.; Buldt, G. *EMBO J.* **1997**, *16*, 1484–1491.
- (12) Kouyama, T.; Nishikawa, T.; Tokuhisa, T.; Okumura, H. *J. Mol. Biol.* **2004**, *335*, 531–546.
- (13) Subramaniam, S.; Greenhalgh, D. A.; Rath, P.; Rothschild, K. J.; Khorana, H. G. *Proc. Natl. Acad. Sci. U.S.A.* **1991**, *88*, 6873–6877.
- (14) Brown, L. S.; Sasaki, J.; Kandori, H.; Maeda, A.; Needleman, R.; Lanyi, J. K. *J. Biol. Chem.* **1995**, *270*, 27122–27126.
- (15) Dioumaev, A. K.; Brown, L. S.; Needleman, R.; Lanyi, J. K. *Biochemistry* **1998**, *37*, 9889–9893.
- (16) Richter, H. T.; Needleman, R.; Kandori, H.; Maeda, A.; Lanyi, J. K. *Biochemistry* **1996**, *35*, 15461–15466.
- (17) Chizhov, I.; Engelhard, M.; Chernavskii, D. S.; Zubov, B.; Hess, B. *Biophys. J.* **1992**, *61*, 1001–1006.
- (18) Ludmann, K.; Gergely, C.; Váró, G. *Biophys. J.* **1998**, *75*, 3110–3119.
- (19) Zscherp, C.; Schlesinger, R.; Heberle, J. *Biochem. Biophys. Res. Commun.* **2001**, *283*, 57–63.

occur in the wild-type photocycle. However, the site-directed mutagenesis data must be interpreted with caution because the coupling between the retinal binding site and the PRG demonstrated by experiments²⁰ and theory³ could lead to structural changes of the active site in the mutants and a PT mechanism different from that in the wild-type.

Because it is difficult to separate O from the previous N intermediate, and the O state does not accumulate in significant amounts in the wild-type photocycle,²¹ a crystal structure of the wild-type O state has not been solved. Two putative O-like structures have been proposed: for the chloride pumping Asp85Ser mutant (pdb code: 1JV7),²² and for the acid-blue bacteriorhodopsin (pdb code: 1X0I).²³ The assignment of these structural models to an O-like state was based on the main characteristics of the O state: neutral Asp85, optical absorption maximum red-shifted relative to the wild-type bR resting state, and all-*trans* retinal. Compared to the bR resting state,²⁴ both O-like structures^{22,23} exhibit significant conformational differences at the extracellular side of the protein (Figure 2a). However, it is unclear to what extent differences in the crystal packing (the space-group is *C222*₁ for the Asp85Ser mutant, and *P622* for acid-blue bacteriorhodopsin) contribute to the structural differences among the two models.

A recent EPR study²⁵ using site-directed spin labels illustrates the specific structural changes (relative to the bR resting state) at the extracellular side of the O state of the Glu204Gln mutant, which is thought to be almost identical to the O state of the wild-type bacteriorhodopsin.²⁵ The changes include an inward movement of helix D toward the proton channel and an increase of the distance between helices C and F (Figure 2a). The structural characteristics of the O state protein in the purple membrane environment are better reproduced²⁵ by the acid-blue bacteriorhodopsin²³ than by Asp85Ser.²² Although the EPR study is a valuable test of the protein conformation in O,²⁵ it does not provide direct information about essential details of structural elements important for PT.

Structural details of the protein groups close to Asp85 can have a significant effect on the energetics of the Asp85 deprotonation. Previous QM/MM computations demonstrated that the proton affinity of Asp85 is greatly influenced by hydrogen bonding water molecules from the active site.²⁶ Water molecules hydrogen bonding to Asp85 and Asp212 could thus be an important determinant of the energy difference between the O and the putative [O] intermediates.

Water molecules close to the retinal Schiff base may also be important for the structure and spectroscopic properties of the active site. Of the three active site water molecules observed in the bR resting state,^{24,27–29} only one is present in the O-like structural models of Asp85Ser mutant of ref 22 and acid-blue

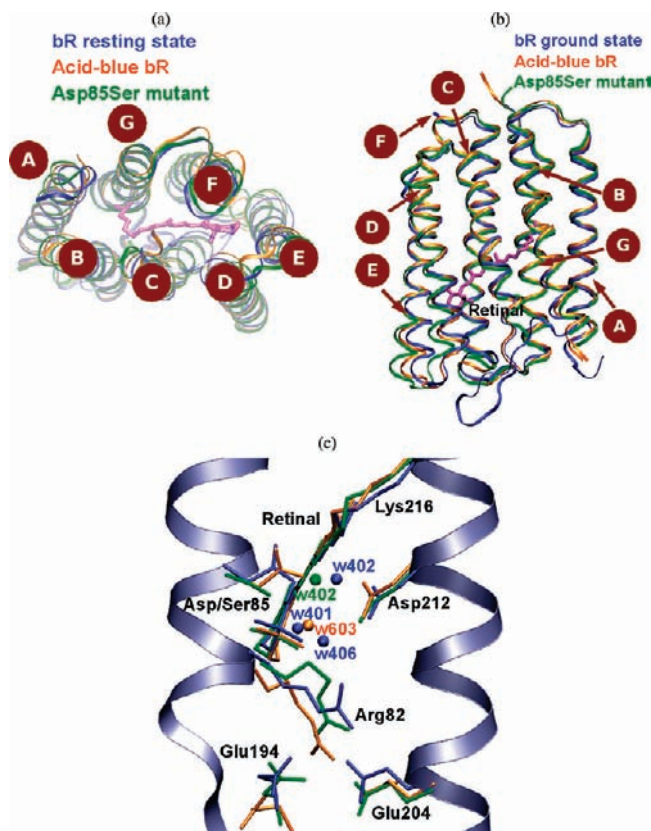


Figure 2. Protein conformational changes in O relative to the bR resting state. Comparison of the crystal structures of the acid-blue bacteriorhodopsin,²³ the Asp85Ser mutant,²² and the bR resting state²⁴ with a view from the cytoplasmic side (a) and parallel to the membrane normal (b). (c) Detailed view of the retinal, water molecules, and amino acids important for the last proton-transfer step. The following color codes are used: orange, acid-blue bR; green, Asp85Ser mutant; blue, bR resting state.

bacteriorhodopsin of ref 23. It is unclear whether indeed there is only one water molecule in the O state active site, or the absence of additional water molecules is due to water molecules being disordered.

Due to the proximity between Asp85/Asp212 and the retinal Schiff base (Figure 2c), the geometry of the retinal could also influence PT. Resonance Raman studies have suggested that the O state retinal chromophore is twisted all-*trans*, with a magnitude of twisting similar to that of the K state structure.³⁰ The retinal twist relaxes as the photocycle finishes with the O → bR transition. Both O-like structures^{22,23} have all-*trans* retinal, possibly mixed with structurally similar 13-*cis*, 15-*syn* retinal.^{22,23}

Given the electrostatic coupling between the active site and the PRG,^{3,20} the geometry of the positively charged Arg82 side chain may be another important determinant of the energetics of PT. In the bR resting state, the guanidinium group of Arg82 is oriented toward the retinal Schiff base region. In contrast, O-like structural models^{22,23} indicate an orientation of the Arg82 side chain toward the extracellular Glu194/Glu204 (Figure 2c). If the guanidinium group of Arg82 indeed adopts an extracellular

(20) Balashov, S. P.; Imasheva, E. S.; Govindjee, R.; Ebrey, T. G. *Biophys. J.* **1996**, *70*, 473–481.

(21) Efremov, R.; Gordeliy, V. I.; Heberle, J.; Buldt, G. *Biophys. J.* **2006**, *91*, 1441–1451.

(22) Rouhani, S.; Cartiailler, J. P.; Facciotti, M. T.; Walian, P.; Needleman, R.; Lanyi, J. K.; Glaeser, R. M.; Luecke, H. *J. Mol. Biol.* **2001**, *313*, 615–628.

(23) Okumura, H.; Murakami, M.; Kouyama, T. *J. Mol. Biol.* **2005**, *351*, 481–495.

(24) Luecke, H.; Schobert, B.; Richter, H. T.; Cartiailler, J. P.; Lanyi, J. K. *J. Mol. Biol.* **1999**, *291*, 899–911.

(25) Chen, D.; Wang, J. M.; Lanyi, J. K. *J. Mol. Biol.* **2007**, *366*, 790–805.

(26) Bondar, A. N.; Suhai, S.; Fischer, S.; Smith, J. C.; Elstner, M. *J. Struct. Biol.* **2007**, *157*, 454–469.

(27) Belrhali, H.; Nollert, P.; Royant, A.; Menzel, C.; Rosenbusch, J. P.; Landau, E. M.; Pebay-Peyroula, E. *Structure* **1999**, *7*, 909–917.

(28) Facciotti, M. T.; Rouhani, S.; Burkard, F. T.; Betancourt, F. M.; Downing, K. H.; Rose, R. B.; McDermott, G.; Glaeser, R. M. *Biophys. J.* **2001**, *81*, 3442–3455.

(29) Matsui, Y.; Sakai, K.; Murakami, M.; Shiro, Y.; Adachi, S.; Okumura, H.; Kouyama, T. *J. Mol. Biol.* **2002**, *324*, 469–481.

(30) Smith, S. O.; Pardo, J. A.; Mulder, P. P. J.; Curry, B.; Lugtenburg, J.; Mathies, R. *Biochemistry* **1983**, *22*, 6141–6148.

orientation in O,^{22,23} it must then return to its original cytoplasmic orientation toward the Schiff base during the O → bR transition. The orientation of Arg82 could also influence the dynamics of the water molecules visiting transiently the protein interior, and whether or not they can reach the active site. It is not known at what stage during the O → bR transition the side chain of Arg82 changes its orientation and whether or not the movement of Arg82 is part of the PT reaction.

In the absence of accurate structural information on the wild-type O intermediate, it is necessary to understand how key structural features affect the deprotonation of Asp85 and to use a comparison between theory and experiments to validate (or invalidate) structural characteristics of the O and the putative [O] intermediates. The structural features assessed here are (i) the role of the retinal geometry, (ii) the number of active site water molecules, and (iii) the geometry of Arg82 side chain. We computed reaction paths using seven different protein models. To validate specific structural characteristics of the models, we computed vertical excitation energies and vibrational spectra. The calculations indicate that an [O] intermediate with deprotonated Asp85/protonated Asp212 could indeed be sampled along the O → bR path. In the O state, retinal is twisted and three water molecules are present in the Schiff base region.

2. Computational Details

2.1. QM/MM Setup. The protein is divided in a quantum mechanics (QM) region that contains the groups involved in the chemical reaction, and the remaining protein is treated with classical molecular mechanics (MM). The total energy is given by

$$E = E_{\text{QM}} + E_{\text{MM}} + E_{\text{QM:MM}} \quad (1)$$

where E_{QM} is the interaction energy between the electrons and the nuclei of the QM region, E_{MM} describes the interactions between the MM atoms, and $E_{\text{QM:MM}}$ represents the interaction energy between the quantum and classical subsystems. The $E_{\text{QM:MM}}$ term can be further decomposed in

$$E_{\text{QM:MM}} = E_{\text{QM:MM}}^{\text{b}} + E_{\text{QM:MM}}^{\text{Coul}} + E_{\text{QM:MM}}^{\text{vdw}} \quad (2)$$

where $E_{\text{QM:MM}}^{\text{b}}$ gives the bonded interactions at the frontier bonds between the QM and MM atoms. $E_{\text{QM:MM}}^{\text{Coul}} + E_{\text{QM:MM}}^{\text{vdw}}$ give the Coulombic and van der Waals interactions, respectively, between the QM and MM particles.

The interaction energy between the classical particles are calculated using the CHARMM molecular force field.³¹ The QM and QM:MM interactions are computed with an approximate DFT method, the self-consistent charge density functional tight binding (SCC-DFTB).³² The implementation of SCC-DFTB in CHARMM has been described in ref 33. SCC-DFTB is ~3 orders of magnitude faster as compared to full DFT methods.³⁴ The accuracy of the SCC-DFTB QM/MM approach for studies of biological systems has been documented in refs 33–41. As a prerequisite for the application to retinal proteins, we demonstrated that SCC-DFTB

is in good agreement with full DFT methods for describing the ground-state properties of protonated Schiff base models (torsional barriers, bond length alternation of the polyene chain, etc.).⁴² In the context of retinal proteins, SCC-DFTB has been applied to study the primary proton transport in bR,^{26,43–46} to refine the rhodopsin crystal structure,⁴⁷ and to generate energy minimized geometries and molecular dynamics (MD) trajectories for excited state calculations.^{48–51} A detailed description of a QM/MM setup for bacteriorhodopsin is given in ref 26.

The combined QM/MM methods^{52–55} have greatly extended the usefulness of theoretical approaches for understanding complex biological systems. We can calculate spectroscopic properties and investigate the energetic details of chemical reactions. QM/MM methods had been applied to study the dynamics of the initial photoisomerization event in bacteriorhodopsin,^{56,57} the structure and spectroscopic properties of the active site in the bR resting state,⁵⁸ and the early intermediate K structure^{45,59,60} and of protonated water clusters in bacteriorhodopsin.^{61,62} The mechanism of the first PT event has been investigated in detail.^{26,43–46,63} Excited state properties of bacteriorhodopsin have been studied with QM/MM approaches,^{48,49,64–67} several of them including polarization effects from the protein environment.^{50,51,56,68}

The QM/MM studies mentioned above and of systems such as carbonic anhydrase II⁶⁹ or Staphylococcal nuclease⁷⁰ have illustrated the strengths and the limitations of the current QM/MM methods.^{55,71–73} The QM/MM computations provided an understanding at an unprecedented level of detail of the possible scenarios for the first PT step and of the energetics associated with the PT paths in bacteriorhodopsin.^{26,43–45,63} However, modern computational models cannot yet reproduce experimental results within the

- (31) Brooks, B. R.; Bruccoleri, R. E.; Olafson, B. D.; States, D. J.; Swaminathan, S.; Karplus, M. *J. Comput. Chem.* **1983**, *4*, 187–217.
 (32) Elstner, M.; Porezag, D.; Jungnickel, G.; Elsner, J.; Haugk, M.; Frauenheim, Th.; Suhai, S.; Seifert, G. *Phys. Rev. B* **1998**, *58*, 7260–7268.
 (33) Cui, Q.; Elstner, M.; Kaxiras, E.; Frauenheim, Th.; Karplus, M. *J. Phys. Chem. B* **2001**, *105*, 569–585.
 (34) Elstner, M. *Theor. Chem. Acc.* **2006**, *116*, 316–325.
 (35) Zhang, X.; Harrison, D. V. T.; Cui, Q. *J. Am. Chem. Soc.* **2002**, *124*, 14871–14878.
 (36) Cui, Q.; Elstner, M.; Karplus, M. *J. Phys. Chem. B* **2002**, *106*, 2721–2740.
 (37) Li, G.; Cui, Q. *J. Am. Chem. Soc.* **2003**, *125*, 15028–15038.
 (38) Li, G.; Zhang, X.; Cui, Q. *J. Phys. Chem. B* **2003**, *107*, 8643–8653.

- (39) Elstner, M.; Cui, Q.; Muni, P.; Kaxiras, E.; Frauenheim, T.; Karplus, M. *J. Comput. Chem.* **2003**, *24*, 565.
 (40) Elstner, M.; Frauenheim, T.; Suhai, S. *THEOCHEM* **2003**, *632*, 29–41.
 (41) Riccardi, D.; König, P.; Prat-Resina, X.; Yu, H.; Elstner, M.; Frauenheim, T.; Cui, Q. *J. Am. Chem. Soc.* **2006**, *128*, 16302–16311.
 (42) Zhou, H.; Tajkhorshid, E.; Frauenheim, T.; Suhai, S.; Elstner, M. *Chem. Phys.* **2002**, *277*, 91–103.
 (43) Bondar, A. N.; Elstner, M.; Suhai, S.; Smith, J. C.; Fischer, S. *Structure* **2004**, *12*, 1–20.
 (44) Bondar, A. N.; Fischer, S.; Smith, J. C.; Elstner, M.; Suhai, S. *J. Am. Chem. Soc.* **2004**, *126*, 14668–14677.
 (45) Bondar, A. N.; Fischer, S.; Suhai, S.; Smith, J. C. *J. Phys. Chem. B* **2005**, *109*, 14786–14788.
 (46) Bondar, A. N.; Baudry, J.; Suhai, S.; Fischer, S.; Smith, J. C. *J. Phys. Chem. B* **2008**, *112*, 14729–14741.
 (47) Okada, T.; Sugihara, M.; Bondar, A. N.; Elstner, M.; Entel, P.; Buss, V. *J. Mol. Biol.* **2004**, *342*, 571–583.
 (48) Wanko, M.; Hoffmann, M.; Strodel, P.; Koslowski, A.; Thiel, W.; Neese, F.; Frauenheim, T.; Elstner, M. *J. Phys. Chem. B* **2005**, *109*, 3606–3615.
 (49) Hoffmann, M.; Wanko, M.; Strodel, P.; Koenig, P.; Frauenheim, T.; Schulten, K.; Thiel, W.; Tajkhorshid, E.; Elstner, M. *J. Am. Chem. Soc.* **2006**, *128*, 10808–10818.
 (50) Wanko, M.; Hoffmann, M.; Frauenheim, T.; Elstner, M. *J. Phys. Chem. B* **2008**, *112*, 11462–11467.
 (51) Wanko, M.; Hoffmann, M.; Fraehmcke, J.; Frauenheim, T.; Elstner, M. *J. Phys. Chem. B* **2008**, *112*, 11468–11478.
 (52) Warshel, A.; Levitt, M. *J. Mol. Biol.* **1976**, *103*, 227–249.
 (53) Field, M.; Bash, P.; Karplus, M. *J. Comput. Chem.* **1990**, *6*, 700–733.
 (54) Gao, J. *Acc. Chem. Res.* **1996**, *29*, 298–305.
 (55) Senn, H. M.; Thiel, W. *Top. Curr. Chem.* **2007**, *268*, 173–290.
 (56) Warshel, A.; Chu, Z. T. *J. Phys. Chem. B* **2001**, *105*, 9857–9871.
 (57) Hayashi, S.; Tajkhorshid, E.; Schulten, K. *Biophys. J.* **2003**, *85*, 1440–1449.
 (58) Hayashi, S.; Ohmine, I. *J. Phys. Chem. B* **2000**, *104*, 10678–10691.
 (59) Hayashi, S.; Tajkhorshid, E.; Schulten, K. *Biophys. J.* **2002**, *83*, 1281–1297.
 (60) Hayashi, S.; Tajkhorshid, E.; Schulten, K. *J. Am. Chem. Soc.* **2004**, *126*, 10516–10517.
 (61) Rousseau, R.; Kleinschmidt, V.; Schmitt, U. W.; Marx, D. *Phys. Chem. Chem. Phys.* **2004**, *6*, 1848–1859.
 (62) Mathias, G.; Marx, D. *Proc. Natl. Acad. Sci. U.S.A.* **2007**, *104*, 6980–6985.

experimental error, e.g., reaction energies within 1 kcal/mol, vibrational frequencies within a few cm^{-1} , and excitation energies within a few nanometers. In spite of the limitations regarding the computation of absolute values, QM/MM computations with a proper treatment of the QM region are reliable for the investigation of relative values.

2.2. Proton Transfer Pathways. To investigate the minimum-energy pathways, we used the Conjugate Peak Refinement (CPR)⁷⁴ method as implemented in the TRAVEL module of CHARMM. The CPR method does not require an explicit definition of the reaction coordinate, but only the energy-minimized reactant and product structures. We investigated the PT scenario in all the structural models described in section 2.5 below. The intermediate path points between the saddle points of the converged CPR pathway were further refined using the Synchronous Chain Minimization (SCM) algorithm⁷⁵ implemented in the TREK module of CHARMM.

2.3. Absorption Shifts. We calculated the excitation energies using a high-level *ab initio* method, Spectroscopy Oriented Configuration Interaction (SORCI)⁷⁶ as implemented in the ORCA quantum chemical package.⁷⁷ SORCI utilizes the division of the first-order interaction space into weakly and strongly perturbing configurations, thereby combining the concepts of the classical multireference configuration interaction and the multireference perturbation theory. While the strongly perturbing configurations are treated variationally, the weakly perturbing configurations are treated with second-order Møller–Plesset perturbation theory. The use of approximate natural orbitals eliminates the problem of choosing a suitable single-particle basis whose quality would affect the final CI result. SORCI gains computational efficiency by use of several thresholds which have been carefully adjusted for the system under study. The thresholds used are $T_{\text{Pre}} = 10^{-3}$, $T_{\text{Nat}} = 10^{-6}$, and $T_{\text{Sel}} = 10^{-6} E_h$ (see Supporting Information of ref 48). Only core orbitals are frozen. We used Ahlrich's SV(P) basis set⁷⁸ which is appropriate for the calculations of the complete chromophore. SORCI has been successfully used to calculate vertical excitation energies for retinal proteins previously, based on

optimized geometries calculated with SCC-DFTB/CHARMM.⁴⁸ The spectral shifts between proteins are described very well; e.g., the shift between bR and SRII could be explained on the basis of structural determinants,⁴⁹ but the absolute excitation energies are overestimated by ~ 0.15 eV.⁴⁸ The overestimation of the excitation energies is partly due to the effects of polarization and dispersion,^{50,51} which are not covered in the widely applied standard QM/MM methods. Since calculations that account for these effects are rather costly, we perform standard QM/MM excited state calculations for all the models and recalculate the excitation energies with polarization effects for several selected models. The effect of protein polarization is described using an interactive polarizable model in which we assign atomic polarizabilities to MM atoms of the protein in addition to the fixed atomic charges.^{50,51}

2.4. Vibrational Frequencies. The standard normal-mode analysis requires diagonalization of the calculated mass-weighted second derivative matrix (Hessian). Since, for a system size of several thousand atoms like bacteriorhodopsin, diagonalization of the full Hessian matrix is computationally demanding, we perform the diagonalization in a reduced basis. To calculate the vibrational frequencies we used the VIBRAN module³¹ of the CHARMM in the QM/MM setup.⁷⁹ The advantage of such an approach consists in an accurate description of the polarization effect by the surrounding MM environment on the QM region.⁷⁹ We computed the vibrational frequencies using SCC-DFTB for the QM region. Recent studies have shown that SCC-DFTB is accurate in reproducing DFT results for vibrational frequencies of various systems.^{80,81} We carried out additional benchmark calculations for the ν_{COOH} mode of an isolated acetic acid molecule and for a strongly hydrogen bonded cyclic acetic acid dimer (see Supporting Information). The calculations demonstrate that SCC-DFTB can accurately reproduce the ν_{COOH} band for the isolated acetic acid molecule (1778 cm^{-1}) compared with the experimental value of 1780 cm^{-1} . For the strongly hydrogen bonded cyclic dimer, the ν_{COOH} band computed with SCC-DFTB is red-shifted by ~ 17 cm^{-1} relative to the experimental value 1725 cm^{-1} (see Table S-1 of the Supporting Information).

2.5. Structural Models. The results reported here are based on three structural models (model-A, model-B, and model-C; see below) distinguished by the protein structure, by the geometry of retinal and the Arg82 side chain, and by the number and location of water molecules close to Asp85/Asp212 (Table 1). To investigate in more detail the role of water molecules, we performed four additional sets of computations in which one or two water molecules not present in the crystal structure^{22,23} were added in the model-A and model-B structures (Table 1). The computations using model-A and model-B with additional water molecules were pursuant to our preliminary observation that the presence of three water molecules in the active site of model-A and model-B leads to an active site geometry similar to that of the bR resting state. For all models used, we denote an O state as neutral Asp85 and negatively charged Asp212 and an [O] state as negatively charged Asp85 and neutral Asp212.

2.5.1. Models with Acid-Blue Form of bR (Model-A). To prepare the model-A we used the crystal structure of acid-blue bacteriorhodopsin.²³ This structure indicates coordinates for one active site water molecule (w603) that bridges Asp85 and Asp212 (Figure 2c). We prepared the following models: (i) model-A^{1w} contains the single active site water molecule w603 from the crystal structure (Figure 3a and b); (ii) model-A^{2w} contains w603 from the crystal structure and w401* (Figure 4a and b); (iii) model-A^{3w} contains water molecules w603, w401*, and w406* (Figure 7a and b).

2.5.2. Models with Asp85Ser Mutant (Model-B). The model-B structural conformers were prepared starting from the Asp85Ser

- (63) Braun-sand, S.; Sharma, P.; Chu, Z.; Pislakov, A. V.; Warshel, A. *Biochim. Biophys. Acta* **2008**, *1777*, 441–452.
- (64) Ren, L.; Martin, C. H.; Wise, K. J.; Gillespie, N. B.; Luecke, H.; Lanyi, J. K.; Spudich, J. L.; Birge, R. R. *Biochemistry* **2001**, *40*, 13906–13914.
- (65) Hayashi, S.; Tajkhorshid, E.; Pebay-Peyroula, E.; Royant, R.; Landau, E.; Navarro, J.; Schulten, K. *J. Phys. Chem. B* **2001**, *105*, 10124–10131.
- (66) Fujimoto, K.; Hasegawa, J.; Hayashi, S.; Kato, S.; Nakatsuji, H. *Chem. Phys. Lett.* **2005**, *414*, 239–242.
- (67) Fujimoto, K.; Hayashi, S.; Hasegawa, J.; Nakatsuji, H. *J. Chem. Theory Comput.* **2007**, *3*, 605–618.
- (68) Matsuura, A.; Sato, H.; Houjou, H.; Saito, S.; Hayashi, T.; Sakurai, M. *J. Comput. Chem.* **2006**, *27*, 1623–1630.
- (69) Schaefer, P.; Riccardi, D.; Cui, Q. *J. Chem. Phys.* **2005**, *123*, 014905–014914.
- (70) Ghosh, N.; Cui, Q. *J. Phys. Chem.* **2008**, *112*, 8387–9397.
- (71) Riccardi, D.; Schaefer, P.; Yang, Y.; Yu, H.; Ghosh, N.; Prat-Resina, X.; Koenig, P.; Li, G.; Xu, D.; Guo, H.; Elstner, M.; Cui, Q. *J. Phys. Chem. B* **2006**, *110*, 6458–6469.
- (72) Elstner, M.; Cui, Q. In *Challenges and Advances in Computational Chemistry and Physics*, Vol. 6; Canuto, S., Ed.; Springer, 2008; Chapter “Combined QM/MM methods for the simulation of condensed phase processes using an approximate DFT approach,” pp 381–405.
- (73) Cui, Q.; Elstner, M. In *Multi-scale Quantum Models for Biocatalysis: Modern Techniques and Applications*; Lee, T.; York, D., Ed.; Springer, in press; Chapter Multi-scale QM/MM methods with Self-Consistent-Charge Density-Functional-Tight-Binding (SCC-DFTB).
- (74) Fischer, S.; Karplus, M. *Chem. Phys. Lett.* **1992**, *194*, 252–261.
- (75) Choi, C.; Elber, R. *J. Chem. Phys.* **1991**, *94*, 751–760.
- (76) Neese, F. *J. Chem. Phys.* **2003**, *119*, 9428–9443.
- (77) Neese, F. *ORCA - An ab initio, density functional and semiempirical program package*, Version 2.3 - Revision 09; Max Planck Institut fuer Strahlenchemie: Muelheim, Germany, 2004.
- (78) Schäfer, A.; Horn, H.; Ahlrichs, R. *J. Chem. Phys.* **1992**, *97*, 2571–2577.

(79) Cui, Q.; Karplus, M. *J. Chem. Phys.* **2000**, *112*, 1133–1149.

(80) Yang, Y.; Yu, H.; York, D.; Cui, Q.; Elstner, M. *J. Phys. Chem. A* **2007**, *111*, 10861–10873.

(81) Yu, H.; Cui, Q. *J. Chem. Phys.* **2007**, *127*, 234504.

Table 1. Details of the End States of the Various O Models^a

| model | starting crystal structure | protonation state | active-site water molecule(s) |
|----------------------------|-------------------------------|----------------------------------------------------------------|-------------------------------|
| Model-A ^{1w} -O | Acid-blue bR ^b | Asp85 ⁰ /Asp212 ⁻¹ /Glu204 ⁻¹ | w603 |
| Model-A ^{1w} -[O] | Acid-blue bR | Asp85 ⁻¹ /Asp212 ⁰ /Glu204 ⁻¹ | w603 |
| Model-A ^{2w} -O | Acid-blue bR | Asp85 ⁰ /Asp212 ⁻¹ /Glu204 ⁻¹ | w603, w401* |
| Model-A ^{2w} -[O] | Acid-blue bR | Asp85 ⁻¹ /Asp212 ⁰ /Glu204 ⁻¹ | w603, w401* |
| Model-A ^{3w} -O | Acid-blue bR | Asp85 ⁰ /Asp212 ⁻¹ /Glu204 ⁻¹ | w603, w401*, w406* |
| Model-A ^{3w} -[O] | Acid-blue bR | Asp85 ⁻¹ /Asp212 ⁰ /Glu204 ⁻¹ | w603, w401*, w406* |
| Model-B ^{1w} -O | Asp85Ser mutant ^c | Asp85 ⁰ /Asp212 ⁻¹ /Glu204 ⁻¹ | w402 |
| Model-B ^{1w} -[O] | Asp85Ser mutant | Asp85 ⁻¹ /Asp212 ⁰ /Glu204 ⁻¹ | w402 |
| Model-B ^{2w} -O | Asp85Ser mutant | Asp85 ⁰ /Asp212 ⁻¹ /Glu204 ⁻¹ | w402, w401* |
| Model-B ^{2w} -[O] | Asp85Ser mutant | Asp85 ⁻¹ /Asp212 ⁰ /Glu204 ⁻¹ | w402, w401* |
| Model-B ^{3w} -O | Asp85Ser mutant | Asp85 ⁰ /Asp212 ⁻¹ /Glu204 ⁻¹ | w402, w401*, w406* |
| Model-B ^{3w} -[O] | Asp85Ser mutant | Asp85 ⁻¹ /Asp212 ⁰ /Glu204 ⁻¹ | w402, w401*, w406* |
| Model-C ^{3w} -O | bR resting state ^d | Asp85 ⁰ /Asp212 ⁻¹ /Glu204 ⁻¹ | w401, w402, w406 |
| Model-C ^{3w} -[O] | bR resting state | Asp85 ⁻¹ /Asp212 ⁰ /Glu204 ⁻¹ | w401, w402, w406 |

^a Extra added waters are marked with asterisk (*). ^b Reference 23. ^c Reference 22. ^d Reference 24.

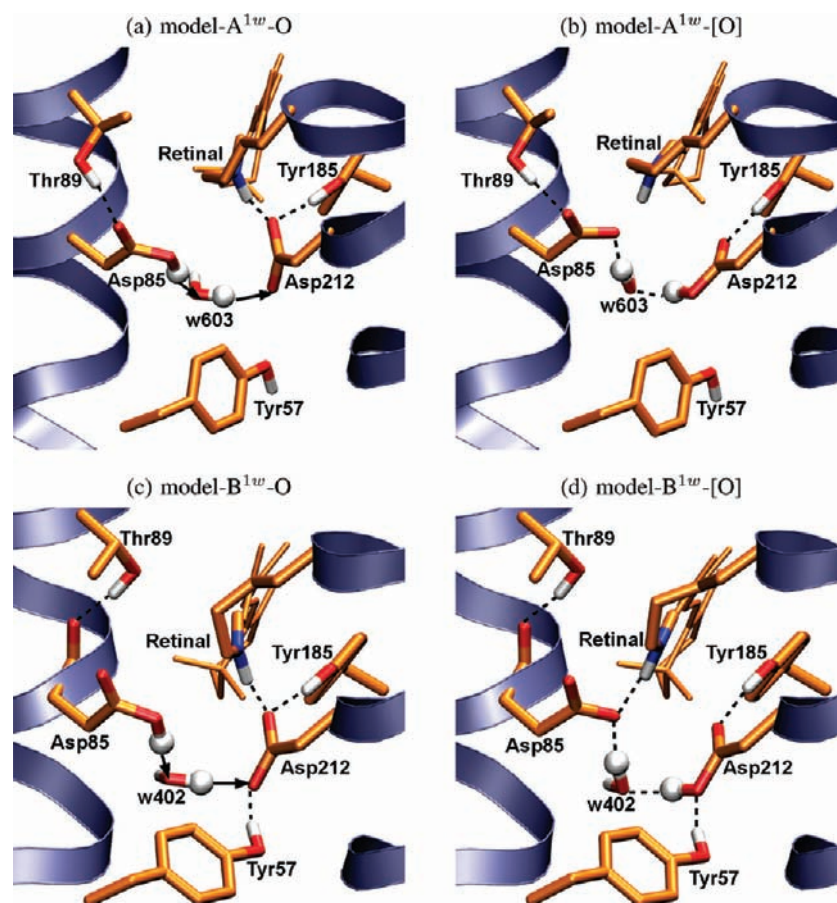


Figure 3. Reactant (O-like) and product ([O]-like) states of proton transfer in the presence of a single water molecule in the active site. The Thr89:Asp85 hydrogen bond is present in model-A^{1w}-O (a) and model-A^{1w}-[O] (b) and absent in model-B^{1w}-O (c) and model-B^{1w}-[O] (d). The Tyr57:Asp212 hydrogen bond is absent in model-A^{1w}-O (a) and model-A^{1w}-[O] (b) and is present in model-B^{1w}-O (c) and model-B^{1w}-[O] (d).

mutant crystal structure.²² The crystal structure has a single water molecule w402 in the active site, and w402 is hydrogen bonded to the Schiff base (N–O distance: 2.7 Å). We mutated Ser85 back into Asp85 and prepared end-conformers for model-B^{1w} (one active site water molecule w402; Figure 3c and d), model-B^{2w} (two active

site water molecules w402 and w401*; Figure 4c and d), model-B^{3w} (three active site water molecules w402, w401*, and w406*; Figure 7c and d).

2.5.3. Model with Resting State of bR (Model-C^{3w}). To assess how the PT energetics is influenced by the protein structure, the

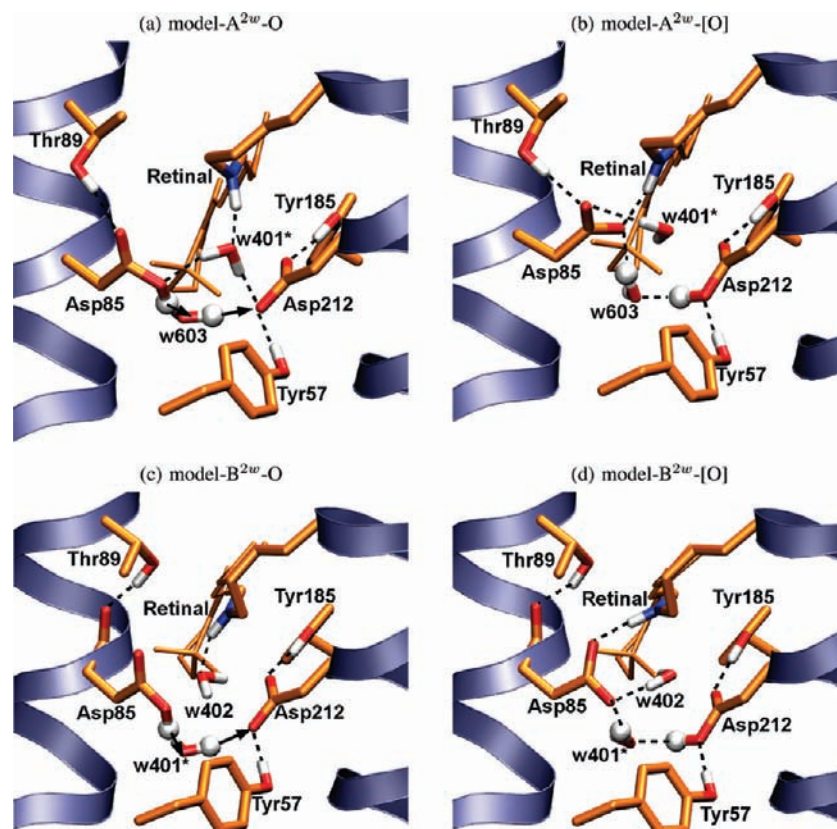


Figure 4. Reactant (O-like) and product ([O]-like) states in the presence of two active site water molecules. The retinal Schiff base hydrogen bonds with w401* in model-A^{2w}-O (a), with w402 in model-B^{2w}-O (c), and with Asp85 in model-A^{2w}-[O] (b) and model-B^{2w}-[O] (d).

Table 2. Activation Energies (ΔE_A , kcal/mol) and Reaction Energies (ΔE_R , kcal/mol) for Proton Transfer from Asp85 to Asp212

| pathway | model | active-site water molecule(s) | member of the proton wire | ΔE_A (kcal/mol) | ΔE_R (kcal/mol) |
|---------|-----------------------|-------------------------------|---------------------------|-------------------------|-------------------------|
| 1a | model-A ^{1w} | w603 | w603 | 5.3 | -0.7 |
| 1b | model-B ^{1w} | w402 | w402 | 4.8 | 3.1 |
| 2a | model-A ^{2w} | w603, w401* | w603 | 3.9 | 0.5 |
| 2b | model-B ^{2w} | w402, w401* | w401* | 10.2 | 5.8 |
| 3a | model-A ^{3w} | w603, w401*, w406* | w603-w406* | 10.9 | 7.4 |
| 3b | model-B ^{2w} | w402, w401*, w406* | w401*-w406* | 6.7 | 3.8 |
| 3c | model-C ^{3w} | w401, w402, w406 | w402 | 13.8 | 3.6 |

orientation of the Arg82 side chain, and the active site water molecules, we performed an additional set of computations starting from the coordinates of the bR resting state (pdb code: 1C3W).²⁴ The active site of model-C^{3w} contains three water molecules: w401, w402, and w406 (Table 1 and Figure 8a and b).

2.6. System Setup. Asp96 and Asp115 are modeled neutral as indicated by experiments.^{1,82} Standard protonation states are used for all remaining titratable protein amino acids. All buried water molecules present in the crystal structures are included in the calculations. The hydrogen atoms are built using the HBUILD utility of CHARMM package.³¹

The QM region consisted of 78 atoms (for model-A^{1w} and model-B^{1w} with one active site water molecule), 81 atoms (for the model-A^{2w} and model-B^{2w} models with 2 active site water molecules), or 84 atoms (for the models with 3 active site water molecules). The QM region comprised Lys216, the complete retinal molecule, the side chains of Asp85, Asp212, and the active site water molecule(s). The boundary between the quantum and classical subsystems was

chosen by adding the link atom at the C _{β} -C _{γ} covalent bond for Lys216 and at the C _{α} -C _{β} covalent bond of Asp85 and Asp212. The QM/MM frontier was treated using a DIV linking scheme.⁸³ In the DIV scheme, the partial charge of the MM host atom is removed from the host and distributed evenly over the remaining MM host atoms.⁸³

To maintain the shape of the protein close to the crystal structure we fixed a part of the protein backbone to the crystal structure coordinates. The remaining part was mobile and consisted of 1578 atoms including the quantum mechanical region and a layer of protein groups and surrounding water molecules that can be related to the PT path. The SCC-DFTB/CHARMM energy minimizations were performed to an rms energy gradient of 10⁻³ kcal/mol·Å.

3. Results

In the following, we discuss QM/MM-optimized models and spectral fingerprint calculations of the O (protonated Asp85 and anionic Asp212) and [O] states (anionic Asp85 and protonated Asp212) and reaction pathways that connect the O and [O] structural models. For each structural model described in Table 1, we computed the vertical excitation energies of O and [O] (Tables 3 and 4) and the vibrational frequencies of the ν_{COOH} band (Table 5).

3.1. Structural Characteristics of End States. 3.1.1. Models with One Active Site Water Molecule. In the crystal structure of the acid-blue bacteriorhodopsin,²³ the distance between Asp85-O ^{δ 1} and Asp212-O ^{δ 2} is shorter (4.2 Å) than that in the bR resting state (5.1 Å) of ref 24. The shorter distance between Asp85 and Asp212 in the O state was explained by either Asp85

(82) Zscherp, C.; Heberle, J. *J. Phys. Chem.* **1997**, *101*, 10542–10547.

(83) König, P. H.; Hoffmann, M.; Frauenheim, Th.; Cui, Q. *J. Phys. Chem. B* **2005**, *109*, 9082–9095.

Table 3. S_1 Excitation Energies (in eV) and Spectral Shifts (in eV) for Various Models^a

| model | state | excitation energy | experimental | | experimental | |
|-----------------------|-------|-------------------|-------------------------|------------|--------------|-------------|
| | | | bR–O shift ^b | bR–O shift | [O]–O shift | [O]–O shift |
| model-A ^{1w} | O | 2.09 | 0.41 | 0.24 | | |
| | [O] | 2.16 | | | 0.07 | 0.04 |
| model-B ^{1w} | O | 2.20 | 0.30 | 0.24 | | |
| | [O] | 2.4 | | | 0.23 | 0.04 |
| model-A ^{2w} | O | 1.92 | 0.58 | 0.24 | | |
| | [O] | 2.06 | | | 0.14 | 0.04 |
| model-B ^{2w} | O | 2.17 | 0.33 | 0.24 | | |
| | [O] | 2.32 | | | 0.16 | 0.04 |
| model-A ^{3w} | O | 1.96 | 0.54 | 0.24 | | |
| | [O] | 2.03 | | | 0.07 | 0.04 |
| model-B ^{3w} | O | 2.15 | 0.35 | 0.24 | | |
| | [O] | 2.29 | | | 0.14 | 0.04 |
| model-C ^{3w} | O | 1.85 | 0.65 | 0.24 | | |
| | [O] | 1.91 | | | 0.06 | 0.04 |

^a Experimental values: br resting state, 2.18 eV;¹⁰⁰ O state, 1.94 eV;¹⁰¹ [O] state, 1.98 eV.¹ ^b Calculated excitation energy of bR: 2.50 eV (Table 4).

Table 4. Effect of Charge Scaling (CS) and Protein Polarization (polar.h) on the Excitation Energies (eV) of Ground-State bR and Various O Models

| model | state | CHARMM | | polar.h |
|-----------------------|--------------|--------|---------|---------|
| | | no CS | with CS | with CS |
| bR | restingstate | 2.50 | 2.41 | 2.24 |
| Model-A ^{3w} | O | 1.96 | 1.94 | 1.92 |
| Model-A ^{3w} | [O] | 2.03 | 1.99 | 2.00 |
| Model-C ^{3w} | O | 1.85 | 1.87 | 1.86 |

Table 5. C=O Stretching Mode (in cm⁻¹) in Various O-Models^a

| model | state | IR mode | O–[O] shift ^b |
|-----------------------|-------|---------|--------------------------|
| Model-A ^{1w} | O | 1690 | 38 |
| | [O] | 1652 | |
| Model-B ^{1w} | O | 1741 | 39 |
| | [O] | 1702 | |
| Model-A ^{2w} | O | 1741 | 39 |
| | [O] | 1702 | |
| Model-B ^{2w} | O | 1758 | 54 |
| | [O] | 1704 | |
| Model-A ^{3w} | O | 1737 | 37 |
| | [O] | 1700 | |
| Model-B ^{3w} | O | 1756 | 37 |
| | [O] | 1719 | |
| Model-C ^{3w} | O | 1728 | 24 |
| | [O] | 1704 | |

^a Experimental values:^{1,82,102} Asp85 in O state, 1752–1756 cm⁻¹; Asp212 in transient O([O]), 1712 cm⁻¹.^{1,82} ^b Experimental O–[O] shift \approx 42 cm⁻¹.

or Asp212 being protonated. In the QM/MM-optimized end-conformers of model-A^{1w}, retinal is twisted. The Schiff base salt bridges to Asp212 in O (model-A^{1w}-O: Figure 3a) and to Asp85 in [O] (model-A^{1w}-[O]: Figure 3b).

In the crystal structure of the Asp85Ser mutant²² water molecule w402 is hydrogen bonded to the Schiff base (O–N distance: 2.7 Å). In the QM/MM-optimized O conformer of model-B^{1w} the Schiff base:w402 hydrogen bond is broken and w402 is displaced toward a more extracellular location, where it forms hydrogen bonds with Asp85 and Asp212 (Figure 3c). Retinal is twisted and the Schiff base salt bridges to Asp212 in O (model-B^{1w}-O: Figure 3c) and to Asp85 in [O] (model-B^{1w}-

Table 6. Torsional Angles (in Degree) around the Double Bonds of the Polyene Chain of Retinal Chromophore in Various O Models

| | C ₇ =C ₈ | C ₁₁ =C ₁₂ | C ₁₃ =C ₁₄ | C ₁₅ =N ₁₆ |
|--------------------------------|--------------------------------|----------------------------------|----------------------------------|----------------------------------|
| bR resting state ²⁴ | 175.1 | 178.9 | –156.9 | –162.7 |
| O Model | C ₇ =C ₈ | C ₁₁ =C ₁₂ | C ₁₃ =C ₁₄ | C ₁₅ =N ₁₆ |
| Model-A ^{1w} | 176.5 | –172.4 | –154.6 | –155.7 |
| Model-B ^{1w} | 172.9 | –174.4 | –168.2 | –170.8 |
| Model-A ^{2w} | 176.2 | –171.2 | –150.0 | –162.9 |
| Model-B ^{2w} | 173.3 | 173.8 | 172.3 | 178.0 |
| Model-A ^{3w} | 174.8 | –170.9 | –151.3 | –165.6 |
| Model-B ^{3w} | 173.2 | –175.4 | –158.0 | –163.8 |
| Model-C ^{3w} | 170.8 | –174.9 | –153.4 | –162.3 |

[O]: Figure 3d). That is, the structures of model-A^{1w} and model-B^{1w} are similar.

The salt bridge between the retinal Schiff base and Asp212 (in O conformer) and between the retinal Schiff base and Asp85 (in [O] conformer) leads to a significant blue-shift of the [O]–O excitation energies relative to the experimental [O]–O shift 0.04 eV.¹ The absolute excitation energies in model-A^{1w} are 2.09 eV for O and 2.16 eV for [O]. In model-B^{1w}, the excitation energies are 2.20 eV for O and 2.44 eV for [O] (Table 3). The differences between the excitation energies of model-A^{1w} and model-B^{1w} models can be explained by differences in the hydrogen bonding pattern of the active site. In model-A^{1w} conformers there is a hydrogen bond between Thr89:Asp85, whereas the Tyr57:Asp212 hydrogen bond is absent (Figure 3a–b). In contrast, in model-B^{1w} conformers, the Thr89:Asp85 hydrogen bond is not present, but the Tyr57:Asp212 hydrogen bond is present (Figure 3c and d). There are also differences in the retinal geometry (see below).

The pattern of hydrogen bonding in the active site may be particularly important for the vibrational frequencies of the ν_{COOH} band. The ν_{COOH} vibrational frequencies of both model-A^{1w}-O (1690 cm⁻¹) and model-A^{1w}-[O] (1652 cm⁻¹) conformers are largely red-shifted by \sim 60 cm⁻¹ relative to the experimental values (Table 5), indicating a more distorted hydrogen bonded network than expected from experiments. The red-shift in the vibrational frequencies computed with model-A^{1w} can be attributed (at least in part) to the presence of the Thr89:Asp85 hydrogen bond. The ν_{COOH} vibrational frequencies computed with model-B^{1w}-O and model-B^{1w}-[O] conformers in which the Thr89:Asp85 hydrogen bond is absent, are within \sim 10 cm⁻¹ from the experimental values (Table 5). Moreover, the O–[O] shifts computed with model-A^{1w} (38 cm⁻¹) and model-B^{1w} (39 cm⁻¹) are in good agreement with the experimental value 42 cm⁻¹.

3.1.2. Models with Two Active Site Water Molecules. In model-A^{2w} prepared from the acid-blue bacteriorhodopsin structure,²³ we inserted w401* within hydrogen bonding distance from the retinal Schiff base (Figure 4a and b). The distance between Asp85–O ^{δ 1} and Asp212–O ^{δ 2} is shorter (4.0 Å) than that in the bR resting state (5.1 Å of ref 24). In the O conformer of model-A^{2w} (model-A^{2w}-O), w401* hydrogen bonds with the Schiff base (2.7 Å), with the negatively charged Asp212 (2.8 Å) and with Asp85 (2.7 Å: Figure 4a). As a result, the presence of w401* and w402 in the active site reduces the retinal twist relative to models (model-A^{1w} and model-B^{1w}) in which only one water molecule was present (Table 6). In the [O] conformer of model-A^{2w} (model-A^{2w}-[O]), the hydrogen bond between the Schiff base and w401* is broken, the retinal chain is twisted, and the Schiff base salt bridges to Asp85 (2.8 Å: Figure 4b).

The excitation energy of model-A^{2w}-O (1.92 eV) is close to the experimental value of 1.94 eV, indicating that binding of the retinal Schiff base to water is an important structural determinant of the electronic structure of the chromophore. Due to the salt bridge between the retinal Schiff base and Asp85, the absolute excitation energy of model-A^{2w}-[O] state (2.06 eV) is somewhat blue-shifted relative to the experimental value of 1.98 eV. The [O]–O shift (0.14 eV) of model-A^{2w} is much larger than the experimental value 0.04 eV. Therefore, accurate models of the O and [O] states may be characterized by the retinal Schiff base being hydrogen bonded to water.

Model-B^{2w} was prepared from the Asp85Ser mutant²² by adding water molecule w401* (Figure 4c and d). The O state of model-B^{2w} (model-B^{2w}-O: Figure 4c) has similar structural features of the active site as in model-A^{2w}-O: there is a hydrogen bond between the Schiff base and a water molecule (w402 in model-B^{2w}-O and w401* in model-A^{2w}-O) and a hydrogen bonded chain between Asp85 and Asp212 with the participation of the second active site water molecule (w603 in model-A^{2w}-O and w401* in model-B^{2w}-O).

The higher excitation energy of model-B^{2w}-O (2.17 eV) compared to the excitation energy of model-A^{2w}-O (1.92 eV: Table 3) may result from the shorter distance between the Schiff base and the counterion Asp212 (3.4 Å in model-B^{2w}-O, compared to 4.1 Å in model-A^{2w}-O). The shorter distance in model-B^{2w}-O leads to an increased stabilization of the positive charge at the Schiff base in the ground state and consequently to a larger excitation energy.

In contrast to the models containing a single water molecule in the active site, the presence of two water molecules in model-A^{2w} and model-B^{2w} is associated with the presence of the Tyr57: Asp212 hydrogen bond. The Thr89:Asp85 hydrogen bond is, however, absent in the model-B^{2w} conformers (Figure 4c and d). The absence of the Thr89:Asp85 hydrogen bond in model-B^{2w}-O leads to a higher ν_{COOH} stretching frequency (1758 cm⁻¹) compared to that computed in model-A^{2w}-O (1741 cm⁻¹), which contains the Thr89:Asp85 hydrogen bond (Figure 4a). These observations support the conclusion from the calculations based on models with a single active site water molecule that the Thr89:Asp85 hydrogen bond plays an important role in tuning of the C=O stretch frequency.

3.1.3. Models with Three Active Site Water Molecules. The discussion above indicates that the structural models with one or two water molecules in the active site are not consistent with the experimental spectral fingerprints. The formation of a salt bridge between the retinal Schiff base and either Asp85 or Asp212 leads to a very distorted retinal geometry and to optical excitation energies that are blue-shifted relative to experimental excitation energies of O and [O] states (Table 3).

To further investigate the configuration of the water molecules in the active site and the relationship between the location of water molecules and the optical excitation spectra, we performed the following two sets of MD simulations (for details of the simulation setup see Supporting Information). To investigate the dynamics of the active site water molecules on a time scale longer than accessible with QM/MM MD simulations, in the first set of simulations we performed three MM MD simulations (5 ns each) for model-A^{1w}-O. During these MD simulations w603 moved away from Asp85/Asp212, and the Asp85-O^{δ1}: Asp212-O^{δ2} distance decreased to ~2.7 Å (Figure 5c), i.e., much shorter than the 4.2 Å distance indicated by the crystal structure.²³ We also observed that water molecules moved from the PRG region closer to the active site (Figure 5b). Therefore

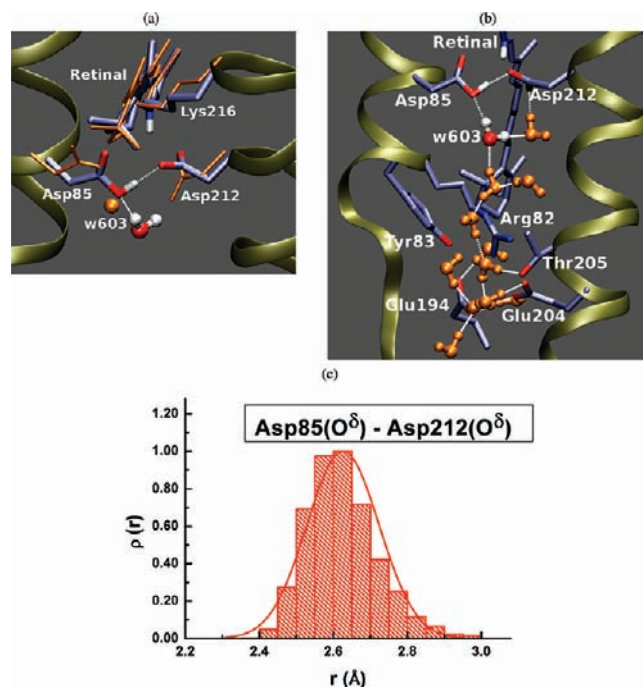


Figure 5. Classical molecular dynamics simulations of model-A^{1w}-O. (a) Comparison of the active site of the crystal structure²³ (orange) with a snapshot from the simulation shows that the essential structural details of the crystal structure are not preserved during the MD simulations. (b) A hydrogen-bonded network forms that connects the proton donor group Asp85 with the PRG Glu194/Glu204. (c) The distance between Asp85 and Asp212 decreases dramatically from 4.2 Å in the crystal structure²³ to ~2.7 Å in the simulations.

one cannot exclude that in O there may be more water molecules in the active site than indicated by the crystal structure.

In the second set of simulations, we performed QM/MM MD for model-A^{3w}-O (Figure 6). We observed that the water molecule w401* that connects the Schiff base to Asp85/Asp212 remained in the active site during the simulations (Figure 6a and b). The Asp85-O^{δ1}:Asp212-O^{δ2} distance (~4.2 Å; Figure 6c) agrees well with the crystal structure²³ value 4.2 Å. The good agreement between the location of water molecule w603 in the crystal structure²³ (Figure 6a) and in the QM/MM MD simulations of model-A^{3w}-O suggests that three water molecules could be accommodated in the crystal structure. Pursuant to these considerations, we performed computations on models with three water molecules as described below.

Model-A^{3w}-O (Figure 7a) has a hydrogen bonded network in which Asp85 and Asp212 are hydrogen bonded via w603 and w406*, and w401* hydrogen bonds with the Schiff base, Asp85, and Asp212. A similar hydrogen bonded network is observed in model-A^{3w}-[O] (Figure 7b). The hydrogen bonding pattern of the active site of model-A^{3w} probably explains the absolute excitation energies of 1.96 and 2.03 eV for O and [O], respectively. The calculated excitation energies are close to the experimental values (1.94 for O and 1.98 for [O]). The calculated shift between the [O] and O models (0.07 eV) is in excellent agreement with the experimental [O]–O shift (Table 3).

The hydrogen bonds between Thr89 and Asp85, and between Asp85 and water molecules (Figure 7a, b), downshift the ν_{COOH} band in model-A^{3w}-O to 1737 cm⁻¹. In model-A^{3w}-[O] in which Asp212 shares hydrogen bonds with Trp86, Tyr57, and Tyr185, the ν_{COOH} band is at 1700 cm⁻¹. Since SCC-DFTB slightly underestimates these carboxyl stretching frequencies when hydrogen bonds are present (see Table S-1 of the Supporting

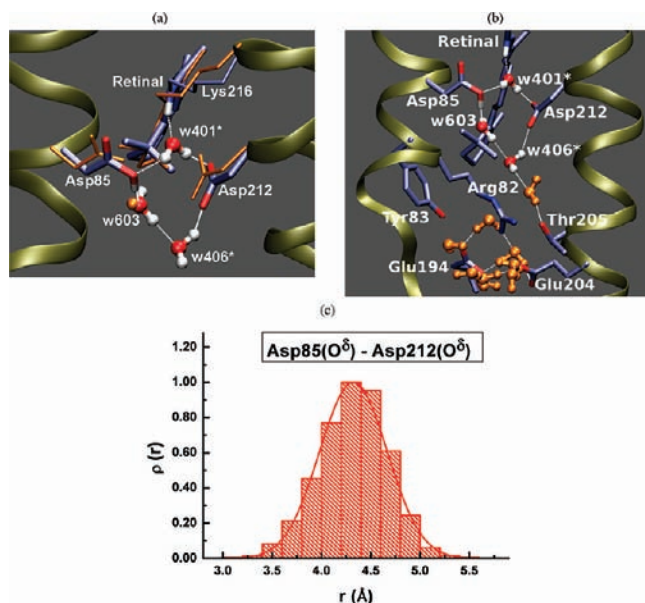


Figure 6. The geometry of the active site indicated in ref 23 is well preserved in MD simulation when three water molecules are present in the active site. The good agreement between the crystal structure and the MD simulations with three water molecules (model-A^{3w}-O) indicate that three water molecules are likely present in the active site. (a) Comparison of the active site geometry of the crystal structure (orange) with a snapshot from the SCC-DFTB MD simulation; (b) A snapshot from MD simulation starting from model-A^{3w}-O; (c) Histogram of the distance between Asp85/Asp212 closest oxygen atoms.

Information), the frequencies computed with model-A^{3w} are in excellent agreement with the experimental values. The O–[O] shift of the ν_{COOH} band (37 cm⁻¹) is also in very good agreement with the experimental O–[O] shift 42 cm⁻¹ (see Table 5).

To estimate the influence of the Thr89:Asp85 hydrogen bond, we constructed a model where this hydrogen bond is absent and Thr89 hydrogen bonds to the backbone carbonyl oxygen of Asp85. The structural model is 1.1 kcal/mol higher in energy than the model in which the Thr89:Asp85 hydrogen bond is present, and the ν_{COOH} frequency (1758 cm⁻¹) is higher than the experimental value of the O state (1752 cm⁻¹).

The active site geometry of the mutant O structures (O and [O] conformers of model-B^{3w}; Figure 7c and d) is similar to that of the acid-blue model (Figure 7a and b). The main difference between model-A^{3w}-O and model-B^{3w}-O is in the absence of the Thr89:Asp85 hydrogen bond in model-B^{3w}-O (Figure 7c). As discussed above for model-A^{3w} conformers without a Thr89:Asp85 hydrogen bond, the absence of this hydrogen bond leads to a blue-shifted ν_{COOH} frequency in model-B^{3w}-O. The ν_{COOH} frequency is at 1756 cm⁻¹ in model-B^{3w}-O and at 1719 cm⁻¹ in model-B^{3w}-[O]. Given that SCC-DFTB slightly underestimates the C=O frequency in strongly hydrogen bonded structures, the ν_{COOH} frequencies computed for model-B^{3w} conformers are too high relative to the experimental values. Moreover, the excitation energies of 2.15 eV for model-B^{3w}-O and 2.29 eV for model-B^{3w}-[O] (Table 3) are somewhat blue-shifted relative to the experimental values. The blue-shift may be due to the missing Thr89:Asp85 hydrogen bond and to the short distance between the retinal C_ε atom and Asp85-O^{δ2} (3.3 Å in model-B^{3w}-O in contrast to 3.8 Å in model-A^{3w}-O). This supports the proposal that the acid-blue model resembles the conformation of the O state in the purple membrane²⁵ and demonstrates that the small spectral shifts calculated between O and [O] for model-A^{3w} are reproducible.

The bR resting state²⁴ has a strong hydrogen bonded network in the active site comprising the Schiff base, the three water molecules w401, w402, and w406, and the two anionic Asp85 and Asp212 groups. The crystal structure of the bR resting state indicates a cytoplasmic orientation of the Arg82 side chain,²⁴ which is preserved in the end states of model-C^{3w} prepared from the bR resting state.²⁴ Both model-C^{3w}-O (Figure 8a) and model-C^{3w}-[O] (Figure 8b) have a twisted all-*trans* retinal. The Thr89:Asp85 hydrogen bond present in the bR resting state crystal structure (2.7 Å) is present in both model-C^{3w}-O and model-C^{3w}-[O] conformers. In model-C^{3w}-O, all water molecules from the active site are hydrogen bonded to Asp212, while, in model-C^{3w}-[O], the water molecules are oriented toward the donor Asp85, and not hydrogen bonded to Asp212.

In contrast to model-A^{3w} and model-B^{3w}, the absolute excitation energies of model-C^{3w}-O (1.85 eV) and model-C^{3w}-[O] (1.91 eV) are red-shifted compared to the experimental values: 1.94 eV for O (by 0.09 eV) and 1.98 eV for [O] (by 0.07 eV), respectively. Qualitatively, this may be expected due to the cytoplasmic orientation of the guanidinium group of Arg82 in model-C^{3w}-O. In model-C^{3w}-O the positive charge of Arg82 is close to the Schiff base and, thus, destabilizes the ground state and decreases the excitation energy relative to the ground state. Although the absolute energies of the O and [O] states are red-shifted relative to the experimental values (see Table 3), the [O]–O shift (0.06 eV) is very close to the experimental value 0.04 eV.

3.2. Proton Transfer Pathways. 3.2.1. PT in Models with One Active Site Water Molecule. In model-A^{1w}, PT begins with the reorientation of the Schiff base away from Asp212. As a result, the salt bridge between the Schiff base and Asp212 is weakened. Simultaneously, Asp85 moves closer to w603, such that the Asp85:w603 hydrogen bond is strengthened. At the transition state ($\lambda = 0.44$ in Figure 9a), the hydrogen bonding distances between Asp85:w603 and between w603:Asp212 are equal to 2.4 Å. After the transition state, the retinal chain becomes more planar, and the reaction continues with the transfer of the proton from Asp85 to w603 and from w603 to Asp212, followed by the formation of the salt bridge between the Schiff base and Asp85.

The rate-limiting barrier of 5.3 kcal/mol ($\lambda = 0.44$; Figure 9a) is dominated by E_{QM} (4.6 kcal/mol) and corresponds to the twisted retinal geometry. The favorable contribution of $E_{\text{QM:MM}}^{\text{nb}}$ (−1.79 kcal/mol) to the rate-limiting barrier corresponds to w603 being strongly hydrogen bonded to Asp85 and Asp212. Following the transition state, a decrease in the retinal twist is associated with a decrease in E_{QM} energy contribution. The $E_{\text{QM:MM}}$ contribution starts rising from −1.79 kcal/mol at the transition state to 5.3 kcal/mol at $\lambda = 1$ and partially compensates for the negative E_{QM} contribution and leads to an overall exothermic energy profile.

We observed a similar PT mechanism in model-B^{1w} (see Supporting Information). Similar to the results for model-A^{1w}, in model-B^{1w} the PT energy barrier is small (~5 kcal/mol; Table 2). Within the accuracy of the methods used, the reaction energy is, however, different in model-A^{1w} and model-B^{1w}, being negative for model-A^{1w} and positive for model-B^{1w}. That is, model-A^{1w} favors the proton on Asp212, whereas model-B^{1w} favors the proton on Asp85. The different reaction energies computed for model-A^{1w} and model-B^{1w} may be related to the Tyr57:Asp212 hydrogen bond, which is present in model-B^{1w} but absent in model-A^{1w}. Test computations with a model-A^{1w} conformer in which the Tyr57:Asp212 hydrogen bond is present

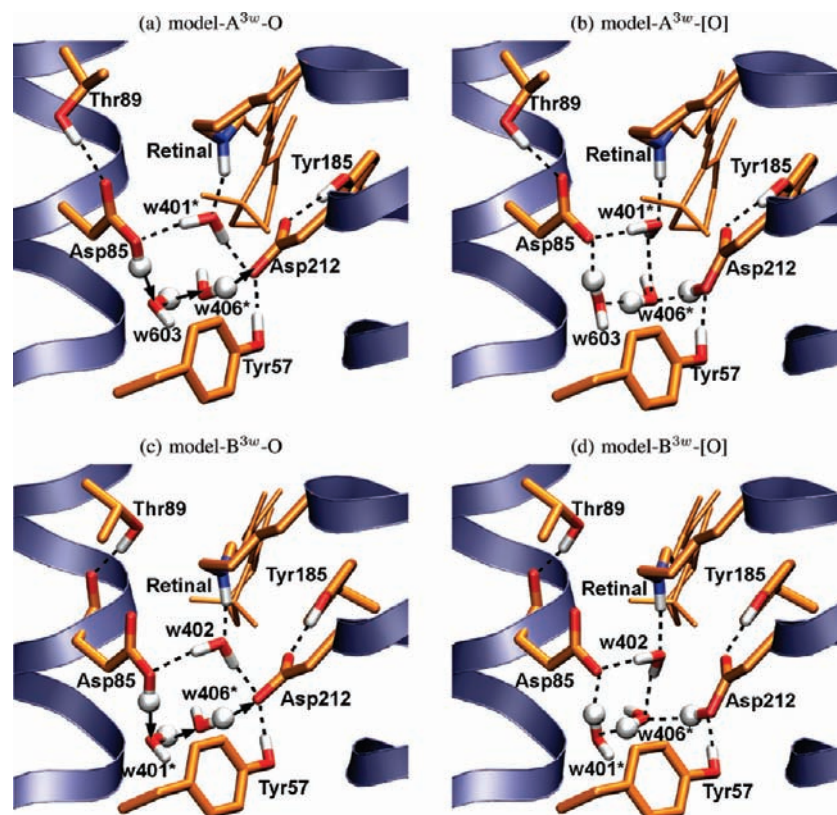


Figure 7. Reactant (O-like) and product ([O]-like) states with three water molecules in the active site. The Schiff base:water hydrogen bond is present regardless of the starting coordinates used for the protein (acid-blue bR^{23} in model- A^{3w} (panels (a) and (b)), and the Asp85Ser mutant in model- B^{3w} (panels (c) and (d)). The Thr89:Asp85 hydrogen bond is present in model- A^{3w} structures (panels (a) and (b)) but absent in model- B^{3w} . The arrows in panels (a) and (c) indicate the PT path.

indicate a reaction energy of 4.5 kcal/mol (Supporting Information) similar to the value 3.1 kcal/mol computed for model- B^{1w} . It is, however, unlikely that either of the model- A^{1w} and model- B^{1w} geometries are a good representation of the active site geometry in O.

3.2.2. PT in Models with Two Active Site Water Molecules.

The mechanism of PT in models with two active site water molecules is similar to that discussed above for models with one active site water molecule (Supporting Information). In the case where two active site water molecules are present, the proton can be transferred either via the water molecule hydrogen bonded to the retinal Schiff base or via the water molecule with the more extracellular location in the active site. When only one water molecule is present in the active site, it acts as an intermediate carrier for the proton.

3.2.3. PT in Models with Three Active Site Water Molecules.

The PT pathway from Asp85 to Asp212 via water molecules w603 and w406* occurs via a concerted mechanism in model- A^{3w} (Figure 9b). The pathway starts with the decrease of the distance between Asp212 and w406* from 2.7 to 2.6 Å, such that the distances between Asp85 and w603, between w603 and w406*, and between w406* and Asp212 become equal to 2.6 Å. At the transition state ($\lambda = 0.47$ in Figure 9b), protons are shared between Asp85-w603 (2.5 Å), between w603-w406* (2.5 Å), and between w406*-Asp212 (2.4 Å). This gives a rate-limiting energy barrier of 10.9 kcal/mol. The reaction then proceeds with PT from Asp85 to w603, from w603 to w406*, and from w406* to Asp212. As a consequence of the transfer, the hydrogen bond between w401* and Asp212 is broken, and a new hydrogen bond is formed between w401* and w406*.

The E_{QM} and $E_{\text{QM:MM}}^{\text{nb}}$ terms contribute 4.9 and 5.0 kcal/mol, respectively, to the rate-limiting barrier (Figure 9b). The increase in $E_{\text{QM:MM}}^{\text{nb}}$ from 5.0 kcal/mol at the transition state to 13.5 kcal/mol at the end of PT (Figure 9b) is due to the increase of the Tyr185:Asp212 and Tyr57:Asp212 hydrogen bonding distances from 2.6 Å in the O state to 2.7 Å in the [O] state.

Due to the close structural similarity between the active sites in model- B^{3w} (Figure 7c) and model- A^{3w} (Figure 7a), the pathway and the energetics for PT in model- B^{3w} are similar to those obtained for model- A^{3w} (see Supporting Information).

PT in model- C^{3w} occurs via water 402 with a large activation barrier of 13.8 kcal/mol and a reaction energy of 3.6 kcal/mol. PT could also occur via water molecules 401 and 406, with a reaction mechanism and associated energetics that are similar to those obtained for model- A^{3w} . To further investigate the influence of the Arg82 orientation on the PT energetics and spectral fingerprints, we used the model- C^{3w} structure to construct a model in which the Arg82 side chain is oriented toward the extracellular side. This conformer is denoted here as model- D^{3w} . The extracellular-oriented Arg82 side chain was taken from the late M-state structure (pdb-code: 1CWQ),⁸⁴ superimposed onto model- C^{3w} . The conformer with an extracellular orientation of the Arg82 side chain is ~ 4 kcal/mol lower in energy than model- C^{3w} . Due to the larger distance between the Schiff base and the positive charge of Arg82 in model- D^{3w} , the absolute excitation energies of O (2.04 eV) and [O] (2.00 eV) states of model- D^{3w} model are blue-shifted relative to

(84) Sass, H. J.; Büldt, G.; Gessenich, R.; Hehn, D.; Neff, D.; Schlesinger, R.; Berendzen, J.; Ormos, P. *Nature (London)* **2000**, *406*, 649–653.

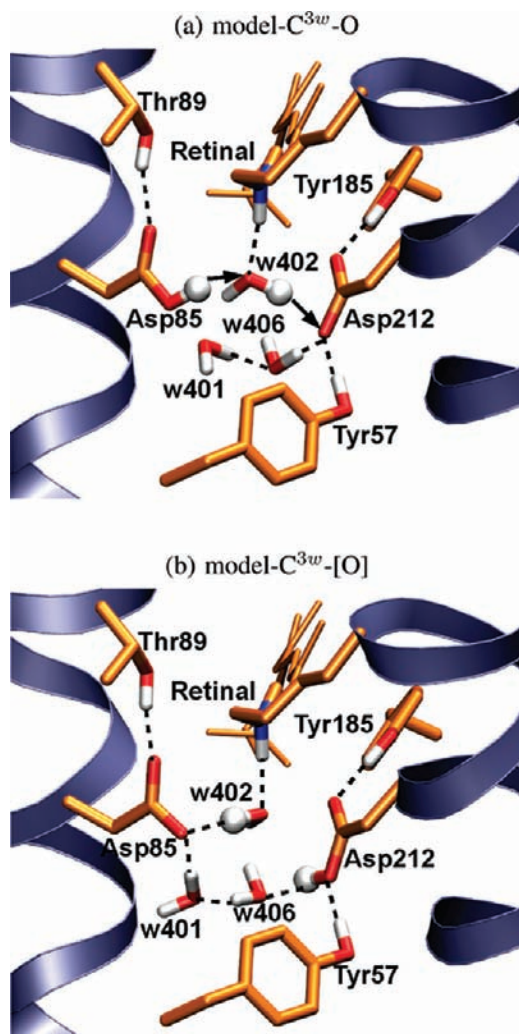


Figure 8. Reactant (O-like) and product ([O]-like) states derived from the bR resting state. In the both model- C^{3w} -O (panel (a)) and model- C^{3w} -[O] (panel (b)) the Schiff base hydrogen bonds with w402, and Thr89 hydrogen bonds with Asp85. The arrows in panel (a) indicate the PT path.

model- C^{3w} (Table 3). The rate-limiting PT barrier computed with model- D^{3w} is 8.0 kcal/mol (see Supporting Information).

3.3. Excited States Calculated by Including Protein Polarization Effects. In a recent study on the influence of the polarizability of the protein/solvent/membrane environment on the calculation of vertical excitation energies, a significant bathochromic shift was found when substituting the fixed CHARMM point-charge model by the polarization model polar.h.^{50,51} The latter combines a polarization-free point charge model with an interactive induced-atomic dipole model to treat polarization within the protein environment explicitly. Furthermore, the screening of solvent-exposed charged groups by the solvent/membrane environment may affect the excitation energy. The Poisson–Boltzmann based charge-scaling scheme of Dinner et al.⁸⁵ represents a simple way to include this effect and is appropriate also for the calculation of excitation energies of retinal proteins.⁵¹

Table 4 shows the effects of the charge scaling and the self-consistent polarization of the chromophore and its environment employing the polar.h model. In the three O-state models, the

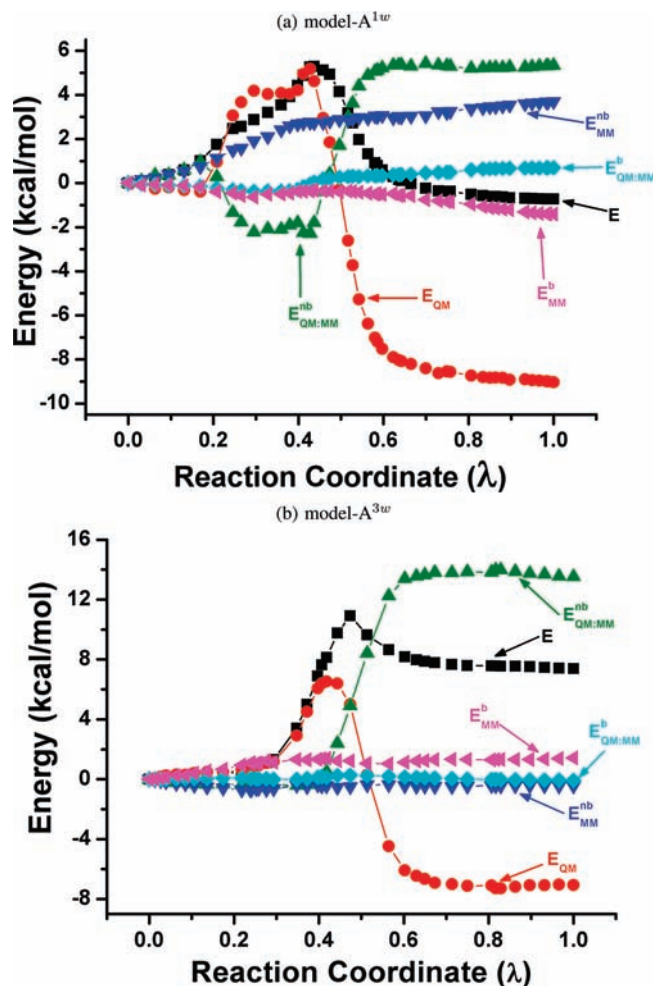


Figure 9. Energy profile of PT from Asp85 to Asp212 calculated with model- A^{1w} (a) and model- A^{3w} (b). The following color codes are used: E (black); E_{QM} (red); $E_{QM:MM}^{nb}$ (green); $E_{QM:MM}^b$ (cyan); E_{MM}^{nb} (blue); and E_{MM}^b (magenta). At $\lambda = 0$ the proton is on Asp85, and at $\lambda = 1.0$ the proton is on Asp212.

net effects are both very small, in contrast to the situation in the bR resting state, where the charge scaling and the polar.h protein electrostatics give rise to bathochromic shifts of 0.09 and 0.17 eV, respectively. The latter is primarily caused by the refined, self-consistent charge distribution in the protein, which includes an electrostatic screening of the doubly charged complex counterion. This screening reduces both the electrostatic interaction with the chromophore and the charge transfer between the protonated Schiff base and the complex counterion. In the O state, only a single counterion is present (Asp212), and these effects are less pronounced.

A minor contribution to the larger bathochromic shift in bR originates from the response of the induced dipoles in the protein to the charge transfer associated with the optical excitation of the chromophore: When neglecting this response, i.e., keeping the ground-state adapted atomic dipoles fixed, the resulting vertical excitation energies are increased by 0.09 and 0.05 eV for the bR resting state and the O state models, respectively. As the polarizability of the protein environment is essentially the same in both cases, the difference correlates with a stronger charge transfer in the bR resting state: The difference-dipole moment $|\Delta S_{-S_0}\mu|$ is 10.8 debye in the bR resting state and 8.4 debye in the model- A^{3w} -O conformer.

(85) Dinner, X.; Lopez, A. R.; Karplus, M. *Theor. Chem. Acc.* **2003**, *109*, 118.

3.4. Retinal Chromophore Configuration. The crystallographic data have shown that the retinal adopts an almost planar geometry in its isolated form,^{86–88} the protein environment induces nonplanarity due to steric and electrostatic interactions.^{89–93,24} Solid-state NMR measurements of the H–C₁₄–C₁₅–H bond along the photocycle indicated that the retinal twist increases by $\sim 17^\circ$ from the bR resting state to early-M.⁹³ FTIR studies on various mutant bR phenotypes suggested that in the O intermediate the retinal is more twisted than in bR and that the twist decreases during the O \rightarrow bR transition.¹⁶

Resonance Raman studies indicated that, in O, the all-*trans* retinal chromophore is twisted;³⁰ the twist is delocalized around the C₇=C₈, C₁₁=C₁₂, and C₁₅=N₁₆ bonds.^{30,92} The dihedral angles of these retinal double bonds for all O-like models investigated here are given in Table 6. The retinal twist is stronger in O than in bR. For example, the dihedral angle C₁₂–C₁₃=C₁₄–C₁₅ is -151.3° in model-A^{3w}-O and -156.9° in the bR resting state (see Table 6). The stronger twist of the retinal chain in the O models than in the bR resting state is consistent with the suggestions from experiments.^{30,16} The maximum twists measured for C₆–C₇=C₈–C₉ and C₁₀–C₁₁=C₁₂–C₁₃ dihedral angles are $\pm 10^\circ$. Larger twists of $\pm 30^\circ$ are observed for the C₁₀–C₁₁=C₁₂–C₁₃ and C₁₄–C₁₅=N₁₆–C _{ϵ} dihedral angles.

Model-A^{1w}-O has large deviations of up to $\sim 20^\circ$ of the twists for C₁₂–C₁₃=C₁₄–C₁₅ and C₁₄–C₁₅=N₁₆–C _{ϵ} dihedral angles relative to the planar all-*trans* value of 180° (see Table 6). Together with the large blue-shift of the excitation energy and the large red-shift of the C=O stretch frequency computed with model-A^{1w}, this distorted retinal geometry indicates that more water molecules may be present in the active site.

Relative to the bR resting state, a large change was observed for C₁₂–C₁₃=C₁₄–C₁₅ and C₁₄–C₁₅=N₁₆–C _{ϵ} dihedral angles for model-A^{3w}-O, model-B^{3w}-O, and model-C^{3w} (Table 6). The significant deviations from planarity of the retinal chain in these structural models is consistent with the experimental findings.^{16,30}

4. Discussion

The aim of this work was to investigate structural models of the O state, and the likelihood that an [O] intermediate characterized by deprotonated Asp85 and protonated Asp212 is sampled during the O \rightarrow bR transition. To address these questions we computed PT, excitation energies, and vibrational frequencies using protein models constructed from the crystal structures of acid-blue bacteriorhodopsin,²³ Asp85Ser mutant,²² and bR resting state.²⁴ The models used in our computations are distinguished not only by the protein structure but also by the number of active site water molecules, orientation of the Arg82 side chain, and the geometry of the retinal polyene chain.

4.1. Role of Active Site Water Molecules. The crystal structures of acid-blue bacteriorhodopsin²³ and Asp85Ser mutant²² indicate only one water molecule is close to the retinal Schiff base. The QM/MM-geometry optimizations reported here indicate that when only one water molecule is present in the active site, local minima in which the Schiff base salt bridges to either Asp85 or Asp212 are energetically preferred. The strong interaction between the Schiff base and Asp85 (or Asp212) leads to optical excitation energies that are blue-shifted relative to the experimental values. That is, the geometries of the active site obtained in the presence of a single water molecule in the active site give spectral fingerprints inconsistent with experiments. Consequently, we constructed additional structural models with two and three water molecules in the active site. The short Asp85:Asp212 distance of 4.2 Å in the acid-blue bacteriorhodopsin crystal structure²³ relative to 5.1 Å in the bR resting state²⁴ is compatible with three water molecules in the active site and does not indicate a direct hydrogen bond between Asp85 and Asp212. The mutual repulsion of the negatively charged Asp85 and Asp212 in the bR resting state is likely reduced in O (where Asp85 is neutral). This lesser repulsion in O allows a decrease of the Asp85:Asp212 distance relative to the bR resting state, as indicated in the acid-blue crystal structure.²³ The QM/MM MD simulations on the acid-blue crystal structure (model-A^{3w}-O) indicate that the active site can accommodate three water molecules.

4.2. Proton Transfer Pathways. We calculated PT pathways between the O and [O] states for all investigated models. Since FTIR studies indicate that the protein environment does not change during the O to [O] transition,¹⁹ we calculated minimal energy pathways; i.e., we neglected the entropic contribution to the energy. The entropic contribution is expected to be more important for the PT to the PRG where the proton is transferred over an ~ 11 Å distance. The activation energy of various PT pathways varies between ~ 4.0 and ~ 11.0 kcal/mol with the exception of the path in model-C^{3w}. The slightly high barrier for model-C^{3w} (13.8 kcal/mol, path 3c) corresponds to the PT via w402 which is strongly hydrogen bonded to the Schiff base. Except for the path computed from the acid-blue structure with one water molecule (path 1a), the energy profiles of the paths (path 1b, 2a-b, and 3a-c; Table 2) indicate small endothermic reaction energies of ~ 4 kcal/mol. The reaction energies and PT barriers obtained for the various PT paths support the idea that a transient [O] state with protonated Asp212 can be sampled during the O \rightarrow bR transition.

4.3. UV–vis Spectral Fingerprints. The calculated optical excitation energies allow us to distinguish between the structural models used. Structural models containing either one or two water molecules in the active site also have a salt bridge between the retinal Schiff base and Asp85 or Asp212. Standard QM/MM computations using these models give excitation energies that are strongly blue-shifted relative to the experimental values. However, since the excitation energy of the bR resting state is also highly overestimated (2.50 eV instead of the experimental value of 2.18 eV), the standard QM/MM calculations are not conclusive.¹⁰⁴ Therefore, we estimated the polarization red-shifts by using a polarizable model for the protein that was developed recently.^{50,51} Polarization effects largely red-shift the absorption of the bR resting state to 2.24 eV (see Table 4). In contrast, in the case of the O state, the effect of polarization on the excitation energies is very small. From the optical spectra computations we conclude that the structural models with only one or two water molecules in the active site are unlikely, since they would

(86) Simmons, C. J.; Liu, R. S. H.; Denny, M.; Seff, K. *Acta Crystallogr. B* **1981**, *37*, 2197–2205.

(87) Hamanaka, T.; Mitsui, T.; Ashida, T.; Kakudo, M. *Acta Crystallogr. B* **1972**, *28*, 214–222.

(88) Santarsiero, B. D.; James, M. N. G.; Mahendran, M.; Childs, R. F. *J. Am. Chem. Soc.* **1990**, *112*, 9416–9418.

(89) Fahmy, K.; Siebert, F.; Grossjean, M. F.; Tavan, P. J. *Mol. Struct.* **1989**, *214*, 257–288.

(90) El-Sayed, M. A.; Lin, C. T.; Mason, W. R. *Proc. Natl. Acad. Sci. U.S.A.* **1989**, *86*, 5376–5379.

(91) Wu, S.; El-Sayed, M. A. *Biophys. J.* **1991**, *60*, 190–197.

(92) Tajkhorshid, E.; Baudry, J.; Schulten, K.; Suhai, S. *Biophys. J.* **2000**, *78*, 683–693.

(93) Lansing, J. C.; Hohwy, M.; Jaroniec, C. P.; Creemers, A. F. L.; Lugtenburg, J.; Herzfeld, J.; Griffin, R. G. *Biochemistry* **2002**, *41*, 431–438.

give a strongly blue-shifted absorption spectrum that is incompatible with experiments. The Arg82 side chain is likely oriented toward the extracellular side, because a cytoplasmic orientation red shifts the absorption spectrum as in the case of model-C^{3w}.

The results in Table 4 indicate that a proper description of the protein electrostatics is required for a quantitative description of the excitation energies for bR resting state and of the shift between the bR resting state and the O state. The theoretical value 0.54 eV for the bR-O shift (model-A^{3w}) is still overestimated (compared to the experimental bR-O shift of 0.24 eV). Although the absolute excitation energies are within 0.06 eV of the experimental absorption maxima. The hypsochromic shift from the O to the transient [O] state is reproduced regardless of the protein electrostatic model employed. The excitation energy for model-C^{3w} is underestimated by 0.08 eV, a value that can be attributed to the cytoplasmic orientation of Arg82, with its positive excess charge closer to the Schiff base, and further away from the retinal β -ionone ring. The excitation energy of model-C^{3w} is significantly blue-shifted when the Arg82 side chain extracellularly oriented as in model-A^{3w}.

The active site of the structure prepared from the Asp85Ser mutant²² with three water molecules (model-B^{3w}) has a rather unusual interaction between the retinal C_i atom and Asp85. Due to the interaction between retinal and Asp85, the excitation energy of model-B^{3w} is strongly blue-shifted relative to the experimental value. From all structural models investigated here, the experimental excitation energies are best reproduced by model-A^{3w}. Model-A^{3w} also reproduces the [O]–O shift in the excitation energy. However, the excitation energies are not sensitive to large scale conformational changes that occur in regions remote from the active site.

4.4. IR Spectral Fingerprints. We used the approximate method SCC-DFTB to compute vibrational frequencies. Using gas-phase models of acetic acid, we benchmarked SCC-DFTB for ν_{COOH} for two extremes scenarios, acetic acid in the gas-phase model and a strongly hydrogen bonded cyclic dimer (see Supporting Information). We discuss all vibrational frequency results with respect to these gas-phase reference calculations. SCC-DFTB reproduces the monomer C=O stretch frequencies extremely well (Table S-1 in the Supporting Information), deviating only by 2 cm⁻¹ relative to the experimental value⁹⁴ 1780 cm⁻¹. The agreement between experiments and theory is worse for the cyclic dimer, where we found a deviation of 17 cm⁻¹. These two test cases are the two extremes of a free C=O group, and of a strongly hydrogen bonded acetate, that mark all hydrogen bonding possibilities in the protein. We expect SCC-DFTB to produce increasingly red-shifted vibrational frequencies with increasing hydrogen bonding interactions. This allows us to assess the O and [O] models with respect to the C=O stretch frequencies. The C=O stretch frequencies do not change when salt bridge forms between the Schiff base and Asp85 or Asp212, being sensitive only to direct hydrogen bonding interactions of the protonated aspartate. The main structural determinants of the aspartate stretching frequencies are the hydrogen bonds with active site water molecules and to nearby groups such as Thr89 (Asp85) and Tyr57 (Asp212). For most models, the [O] state vibrational frequencies are in good agreement with the experimental values, whereas the O state frequencies vary significantly, mostly due to the Thr89:Asp85 hydrogen bond.

In the models used in our study, we find that Thr89 hydrogen bonds either to the side chain (in model-A structures derived from acid-blue bacteriorhodopsin)²³ or to the backbone of Asp85 (in model-B structures derived from the Asp85Ser mutant).²² Since the presence of the Thr89:Asp85 hydrogen bond red-shifts the C=O frequency by ~ 20 cm⁻¹, the hydrogen bonding of Thr89 (to the carboxyl or to the carbonyl group of Asp85) is clearly distinguished by the vibrational frequency of Asp85. Based on the critical examination of the SCC-DFTB performance for vibrational frequencies, we expect slightly red-shifted frequencies compared to experiment (1752–1756 cm⁻¹). Therefore, computation of vibrational frequencies further supports the model-A^{3w} with the Thr89:Asp85 hydrogen bond. The shift in the ν_{COOH} band for O and [O] states is best reproduced in models with three active site water molecules.

Resonance Raman studies have indicated that the all-*trans* retinal in the O intermediate has a twisted geometry.^{30,95} The O-like models used in our study agree with the experimental findings,³⁰ indicating a larger twist in the retinal geometry in O as compared to bR. The retinal twist occurs during the optimization of model-C^{3w}. To obtain a twist of the retinal chain it is sufficient to start from the bR resting state and change the protonation states to correspond to the O state.

5. Conclusion

We modeled the active site of the O state structure using QM/MM simulations. Our investigation of the details of the hydrogen bonded network in the active site indicates that water molecules not present in acid-blue bacteriorhodopsin,²³ Asp85Ser mutant²² crystal structures, have an important structural role in the O intermediate. The geometric and the electronic (absorption energies) features of the retinal chromophore depend on the number of water molecules in the active site. Although the crystal structures of the Asp85Ser mutant²² and of the acid-blue bacteriorhodopsin²³ indicate only one active site water molecule, the geometry of active site from the crystal structure of acid-blue bacteriorhodopsin²³ is best preserved in computations when three water molecules are present in the active site. Similar to the bR resting state,²⁴ in O, the Schiff base hydrogen bonds directly to a water molecule. The absence of the water molecule hydrogen bonding to the Schiff base would lead to large blue-shifts in the optical absorption spectra. The Thr89:Asp85 hydrogen bond present in the first half of the photocycle^{12,24,26,43,96–98} is absent in some of the O state models investigated here. The agreement between the experimental and the calculated C=O stretch frequency computed with the acid-blue bacteriorhodopsin structure in which the Thr89:Asp85 hydrogen bond is present indicates that the Thr89:Asp85 hydrogen bond is indeed a characteristic of the O state. Consistent with the findings of Resonance Raman experiments, all O state models indicate an increased twist of the retinal chromophore with respect to the bR resting state.

To summarize, we propose that the O state model derived from the acid-blue bacteriorhodopsin structure²³ with three active site water molecules, a hydrogen bond between Thr89

(94) Dioumaev, A. K. *Biochemistry (Moscow)* **2001**, *66*, 1269–1276.

(95) Smith, S. O.; Lugtenburg, J.; Mathies, R. J. *Membr. Biol.* **1985**, *85*, 95–109.

(96) Schobert, B.; Cupp-Vickery, J.; Hornak, V.; Smith, S.; Lanyi, J. K. *J. Mol. Biol.* **2002**, *321*, 715–726.

(97) Lanyi, J. K.; Schobert, B. *J. Mol. Biol.* **2002**, *321*, 727–737.

(98) Kandori, H.; Yamazaki, Y.; Shichida, Y.; Raap, J.; Lugtenburg, J.; Belenky, M.; Herzfeld, J. *Proc. Natl. Acad. Sci. U.S.A.* **2001**, *98*, 1571–1576.

and Asp85, and twisted retinal is a good representation of the wild-type O state.

Standard QM/MM excitation energy calculations seem to be insufficient for describing the change in absorption energies between the bR and O state since protein polarization effects are very different in magnitude in these two states. The dependency of the effect of polarization on the structural models used indicates that the polarization effects must be carefully taken into account in studies of color tuning in rhodopsins.

According to the reaction path calculations, a PT from Asp85 to Asp212 is thermodynamically feasible considering the rate-limiting energy barrier of ~ 15 kcal/mol for the O \rightarrow bR reaction.^{18,99} The putative [O], as an intermediate between O and bR, shows changes in the spectroscopic properties as reported experimentally.^{1,82} The experimental value of the shift in the C=O stretch frequency is ~ 40 cm⁻¹, and that of the shift in the optical excitation energy is ~ 0.04 eV between O and [O] states. These changes are well reproduced by our structural models of the O and [O] states modeled from the crystal structure of ref 23 (model-A^{3w}), thereby confirming the picture

of a transient O-intermediate, which is characterized by a protonated Asp212 and deprotonated Asp85. These results indicate that indeed, the spectral shift observed in ref 1 is likely due to Asp212 becoming protonated. The calculations support the proposal of ref 1 that Asp212 is an intermediate carrier of the proton during the O \rightarrow bR transition.

The mechanism of the complete O \rightarrow bR will be the focus of future work. Several key issues must be addressed. For example, the present computations indicate that, in O, the Arg82 side chain is oriented toward the extracellular side. For the recovery of the bR resting state, the side chain must orient back to its cytoplasmic configuration. The rate-limiting energy barrier, the sequence of events associated with the reorientation of the Arg82 side chain, and whether or not reorientation of Arg82 is part of the last PT step are still open questions that will be addressed in the future using the approach presented here.

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Supporting Information Available: Data of all PT pathways. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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- (99) Subramaniam, S.; Faruqi, A. R.; Oesterhelt, D.; Henderson, R. *Proc. Natl. Acad. Sci. U.S.A.* **1997**, *94*, 1767–1772.
- (100) Birge, R. R.; Zhang, C. F. *J. Chem. Phys.* **1990**, *92*, 7178–7195.
- (101) Váró, G.; Lanyi, J. K. *Biochemistry* **1991**, *30*, 5008–5015.
- (102) Hessling, B.; Souvignier, G.; Gerwert, K. *Biophys. J.* **1993**, *65*, 1929–1941.
- (103) Humphrey, W.; Dalke, A.; Schulten, K. *J. Mol. Graphics* **1996**, *14*, 33–38.
- (104) The absolute excitation energy of the bR resting state is 0.07 eV higher than that calculated before at the SORCI level (2.34 eV).⁴⁸ This difference can be attributed to the large QM region used here (retinal, Asp85, Asp212, and three water molecules), whereas in ref 48 only retinal has been treated with QM.